

POOLPRO™

Operation Manual

MODEL PS9TKB

**MYRON L®
COMPANY**

Water Quality Instrumentation
Accuracy • Reliability • Simplicity

24 July 2019

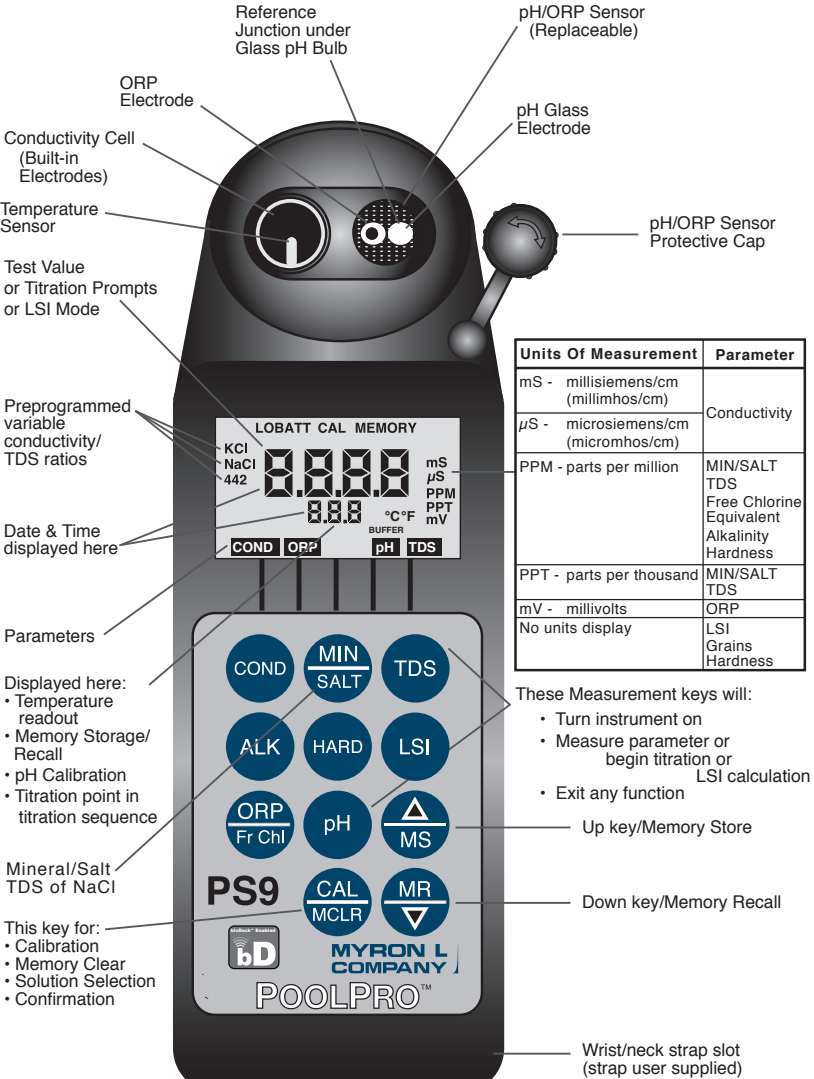
THE PS9 TITRATION KIT (WITH BLUDOCK™ OPTION) COMES WITH EVERYTHING YOU SEE HERE



PLEASE CHECK THE CONTENTS OF YOUR KIT!

1. POOLPRO™ Model PS9-BD
2. Cell Extender Model TKCE
3. Foam-lined hard carry case
4. M'PET 100µL fixed volume pipette Model FVMP-100
5. 12 disposable pipette tips
6. 2oz. bottle of Alkalinity Standard Solution Model ALK-1002OZ
7. 2oz. bottle of Hardness Standard Solution Model HARD-2002OZ
8. 2oz. bottle of pH 4 buffer Model PH42OZ
9. 2oz. bottle of pH 7 buffer Model PH72OZ
10. 2oz. bottle of pH 10 buffer Model PH102OZ
11. 2oz. bottle of pH/ORP Sensor Storage Solution Model SS2OZ
12. 2oz. bottle of NaCl-7500 Model NACL-75002OZ
13. 2oz. bottle of Alkalinity Reagent A1 Model H2SO4-12OZ
14. 2oz. bottle of Hardness Reagent HB Model HBUFF2OZ
15. 2oz. bottle of Hardness Reagent H4 Model EDTA-B-LC2OZ
16. 2oz. bottle of Hardness Reagent H5 Model EDTA-B-HC2OZ
17. 1oz. bottle of ORP ISA Sensor Conditioner Model ORPCOND1OZ
18. 2oz. bottle of KCl-7000 Model KCL-70002OZ
19. 2oz. bottle of 442-3000 Model 442-30002OZ
20. Titration T-plunger Model TPLUNGER

Instrument
Illustration



MODEL PS9
Shown with bluDock™ option installed
For detailed explanations see Table of Contents

I. INTRODUCTION

Thank you for selecting the feature-packed POOLPRO™, one of the Myron L® Company's latest in an increasing line of instruments utilizing advanced microprocessor-based circuitry and SMT manufacturing processes. This circuitry makes the instrument extremely accurate, reliable and very easy to use.

The POOLPRO PS9TKB has been designed to include titration measurements for Alkalinity, Hardness and an LSI Calculator for water balance analysis. The POOLPRO also features Myron L® Company's exclusive Free Chlorine Equivalent (FC^E™) function for making ORP-based free chlorine measurements, as well as optional Bluetooth® wireless data transfer. Other features include a clock with time and date, memory of up to 100 locations with time and date stamp, the ability of the user to adjust the timeout "Auto OFF", and enhanced performance. See Features and Specifications on pages 2-3.

The most exciting feature is data logging with the ability to wirelessly download the memory or stored test data with its corresponding time, date and unit name (requires -BD option or bluDock™ accessory). This feature allows the user to create spreadsheets and graphs with ease, and quickly and accurately manipulate data more effectively. The bluDock's Guardian²™ application will operate on Windows® 7 or later/macOS® 10.13 or later with Bluetooth® capability. The data may be exported to spreadsheet formats such as Microsoft Excel® (.xls and .xlsx) or universal formatting (.csv). The user can also save data in a secure, encrypted format (.mlcx).

Please Note: Although the Myron L® Company has performed extensive testing, we cannot guarantee compatibility of all applications and formats. We suggest testing your application and format for compatibility before relying on it.

For your convenience, a brief set of instructions is provided on the bottom side of your POOLPRO. A Quick Start Guide with abbreviated instructions is also included with the instrument as a quick reference.


Special note ... Conductivity, MIN/SALT, and TDS require mathematical correction to 25°C values (ref. Temperature Compensation, pg. 58). On the left of the POOLPRO's liquid crystal display is shown an indicator of the salt solution characteristic used to model temperature compensation of conductivity and its TDS conversion. The indicator may be KCl, NaCl, or 442™. Selection affects the temperature correction of conductivity and the calculation of TDS from compensated conductivity (ref. Conductivity Conversion to Total Dissolved Solids (TDS), pg. 57). The selection can affect the reported conductivity of hot or cold solutions and will change the reported TDS of a solution. Using KCl for conductivity, NaCl for MIN/SALT, and 442™ (Natural Water characteristic) for TDS is consistent with present industry practice for standardization. This is how your instrument, as shipped from the factory, is set to operate. For use in seawater desalination for example, both the conductivity and TDS may easily be changed to NaCl.

II. FEATURES and SPECIFICATIONS

A. Features

- ORP-based FC^E Free Chlorine measurement; displays as ppm concentration
- Alkalinity and Hardness Conductometric Titrations
- Langelier Saturation Index (LSI) Calculator
- Superior resolution 4 digit LCD displays full 9999 μ S/ppm
- Cond/MIN/SALT/TDS Accuracy of $\pm 1\%$ of READING / $\pm 0.1\%$ at calibration point
- All electrodes are internal for maximum protection
- Improved 4-electrode sensor technology
- Waterproof to 1 meter/3 feet
- Autoranging conductivity/TDS
- Prompts for easy pH calibration
- Factory calibrations stored in microprocessor
- 3 conductivity/TDS solution conversions preprogrammed into microprocessor
- Real Time Clock with Time and Date
- Data Logging with TIME and DATE in memory
- Memory stores 100 readings
- User adjustable timeout “**Auto OFF**”
- Bluetooth[®] wireless download capability with optional bluDock[™] accessory package

B. General Specifications

Display	4 Digit LCD
Dimensions (LxWxH)	196 x 68 x 64 mm/ 7.7 x 2.7 x 2.5 in.
Weight	352 g/12.4 oz.
Case Material	VALOX*
Cond/Res/TDS Cell Material	VALOX*
Cond/TDS Electrodes (4)	316 Stainless Steel
Cond/Res/TDS Cell Capacity	5 ml/0.17 oz.
pH/ORP Sensor Well Capacity	1,2 ml/0.04 oz.
Power	9V Alkaline Battery
Battery Life	>100 Hours/5000 Readings
Operating/Storage Temperature	0-55°C/32-131°F
Protection Ratings	IP67/NEMA 6 (waterproof to 1 meter/3 feet)
 EMI/EMC Ratings	EN61326-1: 2006 + Annex A: 2008 (hand-held devices)
(Conformité Européenne)	CISPR 11: 2003 IEC 61000-4-2: 2001 and, IEC 61000-4-3: 2002

* [™] SABIC Innovative Plastics IP BV

Additional information is available on our website:

www.myronl.com

MADE IN USA

C. Specification Chart

Parameters	Ranges	Resolution	Accuracy	Auto Temperature Compensation	Cond/TDS Ratios Pre-programmed
Conductivity	0-9999 μ S/cm 10-200mS/cm in 5 autoranges	0.01(<100 μ S) 0.1(<1000 μ S) 1(<10mS) 0.01(<100mS) 0.1(<200mS)	$\pm 1\%$ of reading	0-71°C 32-160°F	KCl, NaCl, 442™
TDS	0-9999ppm 10-200ppt in 5 autoranges	0.01(<100ppm) 0.1(<1000ppm) 1(<10ppt) 0.01(<100ppt) 0.1(<200ppt)	$\pm 1\%$ of reading	0-71°C 32-160°F	
MIN/SALT*	0-200ppt in 5 autoranges	0.01(<100ppt) 0.1(<200ppt)	$\pm 1\%$ of reading	0-71°C 32-160°F	
pH	0-14pH	0.01pH	$\pm 0.01\text{pH}^*$	0-71°C 32-160°F	
ORP	$\pm 999\text{mV}$	1mV	$\pm 1\text{mV}^*$		
Free Chlorine Equivalent (FC [™])	0.00-9.99ppm***	0.01ppm	$\pm 0.3\text{ppm}$ $<1.00\text{ppm}^*$ $\pm 0.2\text{ppm}$ $\geq 1.00\text{ppm}^*$	0-71°C 32-160°F	
	350 \leq ORPmV<725 and 0.0 \leq pH<9.9				
	725 \leq ORPmV<825 and 0.0 \leq pH<8.9				
Alkalinity Titration	10-800ppm	0.1ppm			
Hardness Titration	0-1710ppm (0-100 grains)	0.1(<1000 ppm) 1(<1710 ppm) 0.1(<100 grains)			
LSI Calculator	-10 to +10	0.01			
Temperature	0-71°C 32-160°F	0.1°C/°F	$\pm 0.1^\circ\text{C}$		

* NaCl — Sodium chloride ** ± 2 pH in the presence of RF fields ≥ 3 V/m and > 300 MHz

***If either ORP or pH is outside the specified limits, the instrument will display "-Or-".

⚠ WARNING: These products can expose you to chemicals including Di(2-ethylhexyl)phthalate (DEHP), which is known to the State of California to cause cancer and birth defects or other reproductive harm. For more information go to www.P65Warnings.ca.gov

D. Warranty/Service

The Myron L® POOLPRO, excluding the pH/ORP sensor, has a Two (2) Year Limited Warranty. The pH/ORP sensor has a Six (6) Month Limited Warranty for materials and workmanship. If an instrument fails to operate properly, see Troubleshooting Chart, pgs. 50-51. The battery, pH/ORP sensor and cell extender are user-replaceable. For other service, return the instrument prepaid to the Myron L® Company.

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If, in the opinion of the factory, failure was due to materials or workmanship, repair or replacement will be made without charge. A reasonable service charge will be made for diagnosis or repairs due to normal wear, abuse or tampering. This warranty is limited to the repair or replacement of the POOLPRO only. The Myron L® Company assumes no other responsibility or liability.

E. POOLPRO Model

POOLPRO Model	PS9
PARAMETERS	Conductivity/MIN/SALT/TDS/Alkalinity/Hardness/LSI/ORPmV/ Free Chlorine Equivalent (FC ^E) ppm/pH/Temperature

PLEASE NOTE:

Because of our commitment to product improvement, the substance and style of this manual may change. When changes are made, the updated manual is posted for download in PDF format from the Myron L[®] Website: **www.myronl.com**

TABLE OF CONTENTS

PS9TK Contents	i
Instrument Illustration	ii
I. INTRODUCTION	1
II. FEATURES and SPECIFICATIONS	2
A. Features	2
B. General Specifications	2
C. Specification Chart	3
D. Warranty/Service	3
E. POOLPRO Model	4
III. RULES of OPERATION	8
A. Operation	8
B. Characteristics of the Keys	8
C. Operation of the Keys	8
1. Measurement Keys in General	8
2. COND, MIN/SALT and TDS Keys	9
3. Alkalinity, Hardness and LSI Keys	9
4. pH and ORP/FC ^E Keys	10
5. CAL/MCLR Key	10
6. UP or DOWN Keys	10
IV. AFTER USING THE POOLPRO	11
A. Maintenance of the Conductivity Cell	11
B. Maintenance of the pH/ORP Sensor	11
V. SPECIFIC RECOMMENDED MEASURING PROCEDURES	11
A. Parameter Methods	11
B. Titration Methods	11
1. Pipette Instructions	12
2. T-plunger Instructions	13
3. Mixing Solution in Cell	13
C. Measuring Conductivity MIN/SALT & Total Dissolved Solids (TDS)	14
D. Measuring Alkalinity	14
E. Measuring Hardness	16
1. Hardness Unit Selection	16
2. Select EDTA Reagent	16
3. Hardness Titration Procedure	16
F. LSI Calculator Functions	18
1. Measuring LSI	19
2. Hypothetical LSI Calculations	19
G. Measuring pH	20
H. Measuring ORP	21
1. ORP/FC ^E Mode Selection	21
2. Measuring ORP	22
I. Measuring FC ^E	22
1. Prepare for FC ^E Measurement	22
2. FC ^E Flow Method	22
3. FC ^E Immersion Method	23
4. FC ^E Best Practices	24

VI.	SOLUTION SELECTION	24
	A. Why Solution Selection is Available	24
	B. The 3 Solution Types	24
	C. Calibration of Each Solution Type	24
	D. Procedure to Select a Solution	25
VII.	CALIBRATION	26
	A. Calibration Intervals	26
	B. Rules for Calibration of the POOLPRO	26
	1. Calibration Steps	26
	2. Calibration Limits	27
	C. Calibration Procedures	27
	1. Conductivity, MIN/SALT or TDS Calibration	27
	2. Reloading Factory Calibration (Cond, MIN/SALT or TDS)	28
	3. Alkalinity Calibration	28
	4. Hardness Calibration	30
	5. pH Calibration	32
	6. ORP Calibration	34
	7. Temperature Calibration	35
VIII.	CALIBRATION INTERVALS	35
	A. Suggested Intervals	35
	B. Calibration Tracking Records	35
	C. Conductivity, MIN/SALT, TDS Practices	35
	D. pH and ORP/FC ^E Practices	36
IX.	MEMORY	36
	A. Memory Storage	36
	B. Memory Recall	37
	C. Clearing a Record/Memory Clear	37
X.	TIME and DATE	38
	A. Setting TIME	38
	B. Setting DATE	39
	C. DATE FORMAT "US & International (Int)"	40
XI.	TEMPERATURE FORMAT "Centigrade & Fahrenheit"	41
XII.	TOTAL RETURN to FACTORY SETTINGS	41
XIII.	CELL CHECK	42
XIV.	AUTO OFF	43
XV.	bluDock™ Wireless Data Transfer Instructions	44
	A. Software Installation	45
	B. Hardware Setup	45
	C. Memory Stack Download	46
XVI.	CARE and MAINTENANCE	47
	A. Temperature Extremes	47
	B. Battery Replacement	47
	C. pH/ORP Sensor Replacement	47
	D. Cleaning Sensors	47
XVII.	TROUBLESHOOTING	50
XVIII.	ACCESSORIES	52
	A. Conductivity/TDS Standard Solutions	52
	B. Titration Standard Solutions	52

	C. Titration Reagent Solutions	53
	D. pH Buffer Solutions	53
	E. pH Sensor Storage Solution	53
	F. ORP Sensor Conditioner Solution	53
	G. Soft Protective Carry Cases	53
	H. Hard Protective Carry Cases	54
	I. Replacement pH/ORP Sensor	54
	J. Pipette Kit with Replacement Tips	54
	K. Replacement M'Pet Pipette	54
	L. M'Pet Pipette Replacement Tips	54
	M. bluDock™ Wireless Data Transfer Accessory Package	54
XIX.	TEMPERATURE COMPENSATION (Tempco) of Aqueous Solutions	54
	A. Standardized to 25°C	54
	B. Tempco Variation	55
	C. An Example of 2 different solution selections and the resulting compensation	55
	D. A Chart of Comparative Error	56
	E. Other Solutions	56
XX.	CONDUCTIVITY CONVERSION to TOTAL DISSOLVED SOLIDS (TDS)	57
	A. How it's Done	57
	B. Solution Characteristics	57
	C. When does it make a lot of difference?	57
XXI.	TEMPERATURE COMPENSATION (Tempco) and TDS DERIVATION	58
XXII.	pH and ORP/FC ^E	58
	A. pH	58
	B. ORP/Oxidation-Reduction Potential/REDOX	61
	C. FC ^E	62
	1. FC ^E as an Indicator of Sanitizing Strength	62
	2. FC ^E Units	62
XXIII.	ALKALINITY, HARDNESS AND LSI	63
	A. Alkalinity and Hardness Titrations	63
	B. Langelier Saturation Index	63
	C. Hardness Units	64
XXIV.	SOFTWARE VERSION	64
XXV.	GLOSSARY	65

III. RULES of OPERATION

A. Operation

NOTE: The cell extender does not interfere with normal operation. Using the instrument is simple:

- Individual or multiple parameter readings may be obtained by filling individual sensors or entire cell cup area.
- Rinse the conductivity cell and/or pH/ORP sensor well with test solution 3 times and refill. Temperature and/or measurement extremes will require additional rinses for maximum accuracy.
- Press the desired measurement key to start measurement.
- For titrations, user intuitive display prompts guide you through the addition of reagents and measurements.
- Pressing any parameter key again in measurements restarts the 15 second “**Auto off**” timer. Pressing any parameter key again during titrations will cancel the titration in progress.
- Note the value displayed and/or press the **MS** key to store the reading (ref. Memory Storage, pg. 36). It's that simple!

B. Characteristics of the Keys

- Though your POOLPRO has a variety of sophisticated options, it is designed to provide quick, easy, accurate measurements by simply pressing a key.
- All functions are performed one key at a time.
- THERE IS NO “OFF” KEY. If the instrument remains inactive for 15 seconds after you press any parameter key (user adjustable up to 75 seconds), the instrument turns itself off. The instrument turns itself off after 60 seconds of inactivity in CAL mode, 3 minutes of inactivity in each titration screen and 60 seconds of inactivity in each LSI calculator value screen.
- Rarely is it necessary to press and hold a key (as in Procedure to Select a Solution, pg. 25; Conductivity, MIN/SALT, or TDS Calibration, pg. 27; and ORP/FC^E Mode Selection, pg. 21).

C. Operation of the Keys (See Instrument Illustration, pg. ii)

1. Measurement Keys in General

Any of the 8 measurement keys turns on the instrument in the mode selected. The mode is shown at the bottom of the display for COND, RES, TDS, ORP and pH. The ORP mode is indicated by the units displayed, mV for ORP or ppm for free chlorine. Measurement units appear at the right. Pressing any measurement key puts the PS9 in that measurement mode even if you are in a calibration sequence and also serves to cancel a change (ref. Leaving Calibration, pg. 27).

2. COND, MIN/SALT and TDS Keys


These 3 keys are used with solution in the Conductivity Cell.

Precautions:


- While filling cell cup, ensure no air bubbles cling on the cell wall.
- If the proper solution is not selected (KCl, NaCl, 442™), refer to Why Solution Selection is Available, pg. 24 and Procedure to Select a Solution, pg. 25.

a. COND Key


Solution to be tested is introduced into the conductivity cell and a press

of  displays conductivity with units on the right. On the left is shown the solution type selected for conductivity.

b. MIN/SALT Key

A press of  displays Mineral/Salt concentration with PPM or PPT units on the right. On the left the NaCl solution type is shown. An overrange condition will show [- - -] in the display.

c. TDS Key

A press of  displays Total Dissolved Solids with units on the right.


This is a display of the concentration of ionized material calculated from compensated conductivity using the characteristics of a known material. On the left is shown solution type selected for TDS (ref. Solution Selection, pg. 24).

3. Alkalinity, Hardness and LSI Keys


The ALK and HARD keys enter the titration measurement functions. The LSI key allows you to access the LSI Calculator.

NOTE: All titration measurements require the installation of the cell extender.


a. Alkalinity Key

A press of  enters the Alkalinity titration function.

b. Hardness Key

A press of  enters the PS9 in the Hardness titration function.

c. LSI Key


A press of  enters the PS9 in LSI calculator mode, which can be used to measure LSI by pulling the most recent alkalinity, hardness, pH and temperature values or for hypothetical water balance adjustments.

4. pH and ORP/Fr Chl Keys



Measurements are made on solution held in the pH/ORP sensor well (ref. pH and ORP/FC^E, pg. 58). The protective cap is removed and the sensor well is filled and rinsed with the sample enough times to completely replace the storage solution.

After use, the pH/ORP sensor well must be refilled with Myron L[®] Storage Solution, and the protective cap reinstalled securely (ref. Maintenance of the pH/ORP Sensor, pg. 11 and Cleaning Sensors, 2. pH/ORP, pg. 48).


a. pH Key


A press of  displays pH readings. No units are displayed on the right.

b. ORP/Fr Chl Key

In ORP mode, a press of  displays Oxidation-Reduction Potential /REDOX reading in millivolts; "mV" is displayed. When the FC^E mode is activated, a press of  displays the Free Chlorine Equivalent reading in "ppm" alternating with the FC^E predictive ORP reading in "mV".

5. CAL/MCLR Key

While measuring conductivity, TDS, or pH, a press of  allows you to enter the calibration mode. Once in CAL mode, a press of this key accepts the new value. If no more calibration options follow, the instrument returns to measuring (ref. Leaving Calibration, pg. 27).



If  is held down for about 3 seconds, CAL mode is not entered, but "**SEL**" appears to allow Solution Selection (ref. pg. 24) with the Up or Down keys. As in calibration, the **CAL** key is now an "accept" key.

While measuring ORP or Free Chlorine, holding **CAL** down for about 3 seconds allows ORP/FC^E mode selection (ref. pg. 21).

For titrations and LSI Calculator, the **CAL** key steps you through the procedure and accepts values for final calculations.

Once in **CAL** mode, a press of this key accepts the new value. While reviewing stored records, the **MCLR** side of the key is active to allow clearing records (ref. Clearing a Record/Memory Clear, pg. 37).

6. UP or DOWN Keys

While measuring in any parameter, the  or  keys activate the Memory Store and Memory Recall functions. A single press steps the display and holding either key scrolls the value rapidly.

While in calibration or calculator mode, the keys step or scroll the displayed value up or down.

While in Memory Recall, the keys scroll the display up and down through the stack of records (ref. Memory Recall, pg. 37).

IV. AFTER USING THE POOLPRO

A. Maintenance of the Conductivity Cell

Rinse out the cell cup with clean water. Do not scrub the cell. For oily films, squirt in a foaming non-abrasive cleaner and rinse (ref. Cleaning Sensors, pg. 47). Even if a very active chemical discolors the electrodes, this does not affect the accuracy; leave it alone.

B. Maintenance of the pH/ORP Sensor

The sensor well must be kept wet. Before replacing the rubber cap, rinse and fill the sensor well with Myron L[®] pH Sensor Storage Solution. Although not ideal, if Myron L[®] Storage Solution is not available, you can temporarily use a strong KCl solution, a pH 4 Buffer Solution, or a saturated solution of table salt and tap water until pH Storage Solution is obtained. NEVER use distilled water.

If all ORP solutions are reporting approximately the same value, clean ONLY the platinum ORP electrode with an MLC ORP conditioner-soaked cotton swab, being careful not to touch the swab to the glass bulb of the pH sensor.

V. SPECIFIC RECOMMENDED MEASURING PROCEDURES

A. Parameter Methods

For Conductivity, Mineral/Salt and Total Dissolved Solids measurements, if the proper solution is not selected (KCl, NaCl or 442TM), see Solution Selection, pg. 24.

NOTE: After sampling high concentration solutions or temperature extremes, more rinsing may be required. When sampling low conductivity solutions, be sure the pH cap is well seated so that no solution washes into the conductivity cell from around the pH cap.

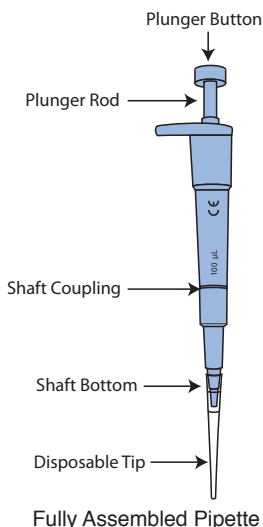
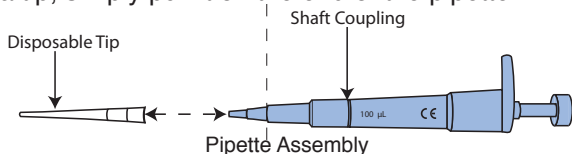
B. Titration Methods

For Alkalinity, Hardness titrations, assemble the required materials before you begin: a pipette with 100 μ L tips, the required reagents, the cell extender, and the T-plunger.

NOTE: The accuracy of titration measurements is affected by your technique. Be careful when removing the cap of the cell extender to add reagents. Flicking the cap or popping it off can cause solution to spill out of the cell extender. Always use a new pipette tip when changing solutions to avoid contamination.

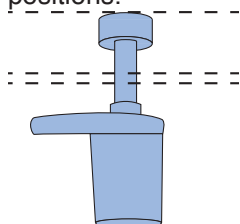
1. Pipette Instructions

Always use a clean tip when changing solutions! To install a tip, press the wide end of the tip on the end of the pipette until it is secure. Avoid contaminating the pipette tips. To remove a tip, simply pull it off the end of the pipette.

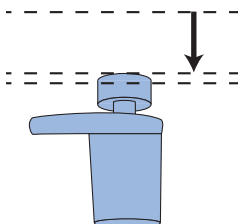


NOTE: Do NOT twist the cap off as this may unscrew the shaft bottom from the pipette.

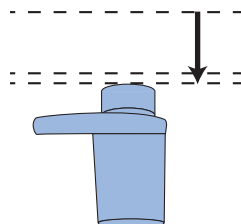
The pipette has 3 positions: REST, FIRST STOP, and SECOND STOP. Practice pushing in and letting up on the pipette to feel the different positions.



Rest Position: The plunger is all the way UP.



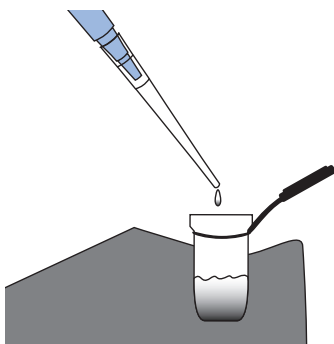
First Stop: The plunger is depressed until resistance is felt.



Second Stop: The plunger is depressed all the way DOWN.

a. To **DRAW** a sample: Grasp the pipette by the shaft with your hand. Rest your thumb on top of the plunger button. Use your thumb to depress the top of the pipette to the **FIRST STOP**. Insert the tip end down into sample until only the tip is submerged, about 2-3mm/1/8". Slowly release the top and let it return to the **REST** position being careful to keep the tip submerged.

b. To **DISPENSE** a sample: With the pipette in the **REST** position, place the tip end over the



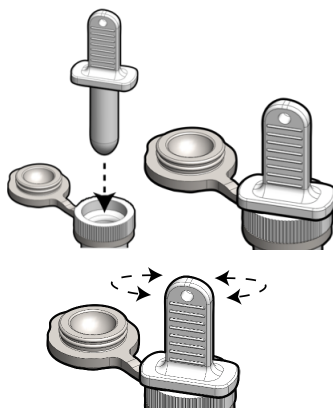
Do not allow the pipette tip to touch the cell wall or the sample.

conductivity cell **being careful not to touch the tip to the existing solution.** Depress the top of the pipette to the **SECOND STOP** (all the way down) being careful to keep the tip over the cell. Release the plunger button and let it return to the **REST** position.

2. T-Plunger Instructions

To **PLUNGE** the cell:

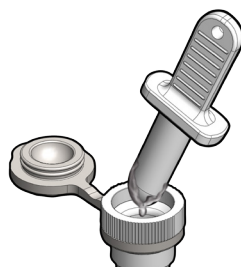
Clean and dry the T-plunger. With the cell extender installed and solution in the cell, insert the tip of the T-plunger in the cell extender until the arms of the T-plunger are flush against the rim of the cell extender. Solution will overflow the cell.



Insert the T-plunger all the way down into the cell extender then rotate it from side to side.

Keeping the arms of the T-plunger flush against the rim of the cell extender, rotate the T-plunger from side to side, allowing the arms of the T-plunger to slide along the rim of the cell extender.

Remove the T-plunger from the cell by lifting it straight up out of the cell, being careful to keep it centered over the mouth of the cell extender. When the T-plunger is completely out of solution, tap the T-plunger on the inside rim of the cell extender to dispel any clinging solution back into the cell.



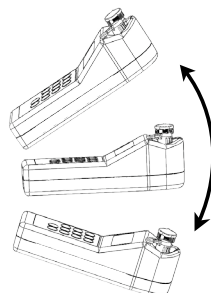
Hold the T-plunger over the cell then tap firmly so that solution is dispelled back into the cell cup.

Remove the T-plunger completely from the cell area.

3. Mixing Solution in Cell

a. Agitate

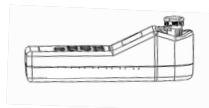
When the display says to agitate ("**AGit**"): Grasp the instrument on both sides of the keypad (keypad facing up) with your hands so you don't accidentally drop or release it. Agitate the solution in the cell by swinging the cell cup area of the PS9 up and down five or more times.



"**AGit**": When "**AGit**" displays, swing the cell end of the PS9 up and down five or more times.

b. Hold

When the display says to hold (“**HOLD**”):
Hold the instrument steady in the horizontal position while the instrument records a measurement.



“**HOLD**”: When “**HOLD**” displays, hold the PS9 still with bottom of case parallel to the ground.

C. Measuring Conductivity Mineral/Salt Concentration & Total Dissolved Solids (TDS)

1. Rinse cell cup 3 times with sample to be measured. (This conditions the temperature compensation network and prepares the cell.)
2. Refill cell cup with sample.
3. Press **COND**, **MIN SALT** or **TDS**.
4. Take reading. A display of [- - -] indicates an overrange condition.

D. Measuring Alkalinity

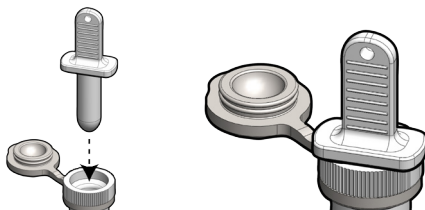
Prepare the materials required for titration: Reagent: A1, cell extender, T-plunger, and 100 μ L pipette with a clean tip installed.

NOTE: If you do not complete each titration point within the 3-minute timeout period, the PS9 exits the Alkalinity function. Start the titration process over.

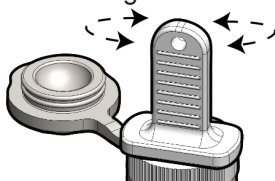
If you make a gross error in your titration method, e.g., you forget to add sample to the cell, “**Err**” will display. Start the titration over.

1. Ensure pH protective cap is secure to avoid contamination.
2. Install the cell extender by pushing the base of the cell extender into the conductivity cell until it is fully seated.
3. Rinse the conductivity cell cup and cell extender 3 times with solution to be measured.

4. Refill cell cup and cell extender with sample solution.
5. Insert T-plunger into open cell extender until the "T" rests flush on the lip of the cell extender. This will cause fluid beyond what is required for titration to overflow the cell extender.



6. Rotate the T-plunger from side to side, allowing the arms of the T-plunger to slide along the rim of the cell extender.



7. Carefully remove the T-plunger so as not to flick or spill remaining sample. Tap the T-plunger on the inside rim of the cell extender to remove sample that may be stuck to the T-plunger.
8. Press **ALK**. "**AL**" alternating with "**tdS**" briefly displays along with the "**PPM**" value of the sample. "**PRES CAL**" will display when the reading stabilizes.
9. Press **CAL** to accept and advance to the first titration prompt. "**Add**" alternating with "**A1**" displays.
10. Using the pipette, add 100 μ L of Reagent: A1, close the cell extender cap securely, press **CAL** to continue.
11. "**AGit t1**" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue. (The number after "t" indicates which titration point you are on. The number "1" here indicates this is the first titration point. Subsequent titration points are sequenced numerically: 2, 3, 4, etc.)
12. "**HOLD**" displays. Keep holding the meter steady until "**Add**" alternating with "**A1**" displays. Open the cell extender cap carefully to avoid spilling, and use the pipette to add 100 μ L of Reagent A1. Close the cell extender cap securely. Press **CAL** to continue.

13. Repeat Steps 11 and 12 until the meter reports the alkalinity result.
14. Note the value for your records or press **MS** to store the value.
The PS9 will automatically power off.

E. Measuring Hardness

1. Hardness Unit Selection

The PS9 offers the ability to set the hardness unit preference to either “**PPM**” CaCO_3 or grains of hardness. To change hardness units, press **HARD**, then press and hold **CAL** down until “**HArd**” and “**SEL**” are displayed. Press **UP** or **DOWN** to toggle between “**PPM**” and grains (no units are displayed for grains). Press **CAL** to accept. This also sets the hardness unit preference for the LSI Calculator simultaneously.

2. Select EDTA Reagent

0-200 ppm- Reagent H4, EDTA-B-LC (optimal for 0-100 ppm)

0-1710 ppm- Reagent H5, EDTA-B-HC (optimal for 100-1710 ppm)

3. Hardness Titration Procedure

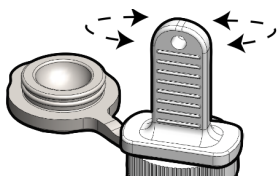
Prepare the materials required for titration: Reagent: HB, Reagent: H4 or Reagent: H5, cell extender, T-plunger, 100 μL pipette with a clean tip installed, and one extra clean tip.

NOTE: If you do not complete each titration point within the 3-minute timeout period, the PS9 exits the Hardness function. Start the titration process over.

1. Ensure pH protective cap is secure to avoid contamination.
2. Install the cell extender by pushing the base of the cell extender into the conductivity cell until it is fully seated.
3. Rinse the conductivity cell cup and cell extender 3 times with solution to be measured.
4. Refill cell cup and cell extender with sample solution.
5. Insert T-plunger into open cell extender until the “T” rests flush on the lip of the cell extender. This will cause fluid beyond what is required for titration to overflow the cell extender.



6. Rotate the T-plunger from side to side, allowing the arms of the T-plunger to slide along the rim of the cell extender.



7. Carefully remove the T-plunger so as not to flick or spill remaining sample. Tap the T-plunger on the inside rim of the cell extender to remove sample that may be stuck to the T-plunger.
8. Press **HARD**. "**CA**" alternating with "**tdS**" displays along with "**PPM**" sample value. "**PRES CAL**" displays when the reading stabilizes.
9. Press **CAL**. "**HC**" or "**LC**" will display; use the **UP** or **DOWN** keys to switch to **HC** for the higher range (0-1710 ppm) or **LC** for the lower range (0-200ppm). Press **CAL** to accept and continue.
10. "**Add**" alternating with "**buFF**" displays. Using the pipette, add 100 μ L of Reagent: HB to the sample, close the cell extender cap securely. Change the pipette tip for adding **EDTA** in following steps. Press **CAL** to continue.
11. "**AGit**" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue.
12. "**HOLD**" displays. Keep holding the meter steady until "**Add**" alternating with "**EdtA**" displays. Open the cell extender cap carefully to avoid spilling, and use the pipette to add 100 μ L of **EDTA-B-HC** or **EDTA-B-LC**, as indicated on the display. Close the cell extender cap securely. Press **CAL** to continue.
13. "**AGit t1**" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue.
(The number after "t" indicates which titration point you are on. The number "1" here indicates this is the first titration point. Subsequent titration points are sequenced numerically: 2, 3, 4, etc.)
14. Repeat Steps 12 and 13 until the meter reports the hardness result.
15. Note the hardness value for your records or press **MS** to store the value. The PS9 will automatically power off.

NOTE: If there is a gross error in your titration method, e.g., you forgot to add sample to the cell, “**rEAd Err**” will alternate with “**rEPt tit**”. Repeat the titration from Step 1 to obtain a reading. If the reading is over range, the display indicates “**Or**”. If the reading is under range, the display indicates “**0.00**”.

F. LSI Calculator Functions

The POOLPRO PS9 features an LSI Calculator that can be used to perform actual LSI or for hypothetical water balance calculations.


The LSI Calculator computes a Langelier Saturation Index value using measured, default, or user-adjusted values for alkalinity, hardness, pH and temperature.

If you have stored alkalinity, hardness or pH and temperature values, the calculator will automatically display the most recent stored value. All other values will display as default. You can adjust any or all of the values displayed to determine the effect of the change(s) on the LSI value.

1. Measuring LSI

To compute the actual saturation index of a solution:


Before you enter the calculator function, you must measure and store values for all water balance variables used by the calculator to compute saturation index.

1. Perform an alkalinity titration of the sample solution (ref. Measuring Alkalinity, pg. 14). Press **MS** to store the reading in memory.
2. Perform a hardness titration of the sample solution (ref. Measuring Hardness, pg. 16). Press **MS** to store the reading in memory.
3. Measure the pH of the sample solution (ref. Measuring pH, pg. 20). Press **MS** to store the reading in memory.
4. Press .
5. The last stored alkalinity value is displayed.
6. Press **CAL** to accept value and advance to the hardness value screen. The last stored hardness value is displayed.
7. Press **CAL** to accept value and advance to the pH value screen. The last stored pH value is displayed.
8. Press **CAL** to accept value and advance to the temperature value screen. The last stored temperature value (taken from the last stored hardness or alkalinity titration) is displayed.
9. Press **CAL** to accept and calculate LSI. The saturation index value will display. Press **MS** to store the reading. The PS9 will automatically power off after the period of inactivity defined in the AUTO OFF setting (ref. AUTO OFF, pg. 43).

If you want to modify any of the input values and recalculate LSI based on those changes, press **CAL** again and repeat Steps 5-9 using the **UP** and **DOWN** Keys to change the values. If you want to change hardness units (ppm/grains), you must do so in the hardness parameter. See Hardness Unit Selection, pg. 16.

2. Hypothetical LSI Calculations


To compute saturation index using hypothetical alkalinity, hardness, pH or temperature values:

1. Press .
2. Either the last stored value or the default value of “120” is displayed.

3. Press the **UP** or **DOWN** keys to adjust the alkalinity value or leave as displayed.
4. Press **CAL** to accept and advance to the hardness value screen. Either the last stored value or the default value of “**166**” is displayed.
5. Press the **UP** or **DOWN** keys to adjust the hardness value or leave as displayed.
6. Press **CAL** to accept hardness value and advance to the pH value screen. Either the last stored value or the default value of “**7.20**” is displayed.
7. Press **UP** or **DOWN** to adjust the pH value or leave as displayed.
8. Press **CAL** to accept pH value and advance to the temperature value screen. Either the last stored value (taken from the last stored hardness or alkalinity titration) or the default value of “**25 °C**” is displayed.
9. Press **UP** or **DOWN** to adjust the temperature value or leave as displayed.
10. Press **CAL** to accept and calculate LSI. A saturation index value will display.

If you want to modify any of the input values and recalculate LSI based on those changes, press **CAL** again and repeat Steps 3-10. If you want to change hardness units, you must do so in the hardness parameter. See Hardness Unit Selection, pg. 16.

G. Measuring pH

1. Remove protective cap by rotating while grasping and pulling up.
2. Rinse pH/ORP sensor well and conductivity cell cup 3 times with sample to be measured. Shake out each sample to remove any residual liquid.
3. Refill both sensor well and cell cup with sample.
4. Press .
5. Note value displayed.
6. **IMPORTANT:** The sensor well must be kept wet. Before replacing the rubber cap, rinse and fill the sensor well with Myron L® pH Sensor Storage Solution.

H. Measuring ORP

The POOLPRO features the ability to measure the activity of oxidizing or reducing chemicals in solution as ORP mV. The instrument also includes an innovative Free Chlorine Equivalent (FC^E) feature (Measuring Free Chlorine Using FC^E, pg. 22) that uses ORP and pH to measure free available chlorine (FAC) concentration in ppm. ORP mV and ppm of free available chlorine (FAC) are the two most commonly used sanitizer units of measure in water quality management.

1. ORP / FC^E Mode Selection

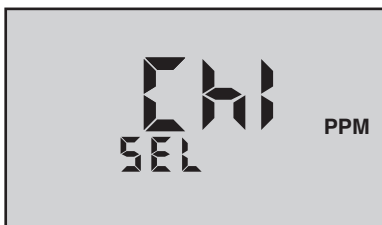
The POOLPRO allows the user to choose between measuring oxidizing sanitizers using either ORP mV or as parts per million (ppm) of equivalent free chlorine. Use ORP to directly measure the oxidizing power of all sanitizers like ozone, bromine, peracetic acid or chlorine. Use FC^E to measure the strength of oxidizing sanitizers as ppm of equivalent free chlorine. To select between ORP and Free Chlorine Equivalent modes:

1. Press .
2. Press and hold for approximately 3 seconds.

The current preference for ORP units of measure is displayed. Factory setting for this preference is ORP mV.




3. Press the or keys to toggle between mV (standard ORP mode) and FC^E ppm. The setting chosen is displayed.



4. Press any parameter key to exit ORP unit preference selection or let the PS9 time out. ORP unit preference will be saved.

2. Measuring ORP

1. Ensure the PS9 is in ORP/FC^E mode (ref. ORP/FC^E Mode Selection, pg. 21).
2. Remove protective cap by rotating while grasping and pulling up.
3. Rinse sensor well and cell cup 3 times with sample to be measured. Shake out each sample to remove any residual liquid.
4. Refill both sensor well and cell cup with sample.
5. Press .
6. Take reading.
7. Press **MS** to store reading in memory, if desired.

IMPORTANT: The sensor well must be kept wet. Before replacing the rubber cap, rinse and fill the sensor well with Myron L[®] pH Sensor Storage Solution.

1. Measuring Free Chlorine Using FC^E



The FC^E function can be used to measure flowing solution, bodies of water and discrete samples. The measurement techniques are unique to each type of sample. For best results, use the FC^E Flow Method or the FC^E Immersion Method. For discrete samples, follow the FC^E Flow method, but instead of a running stream, rinse the conductivity cell cup and pH sensor well three times then fill with the sample to be tested. Read through Section 4: FC^E Best Practices before you begin.

1. Prepare for FC^E Measurement

1. For ease of measurement, set the instrument's Auto OFF feature to 75 sec (ref. AUTO OFF, pg. 43).
2. Ensure the FC^E mode has been activated (ref. ORP/FC^E Mode Selection, pg. 21).
3. Remove protective cap from the pH/ORP sensor by rotating while grasping and pulling up.


2. FC^E Flow Method

1. Empty the pH/ORP sensor well of all storage solution.
2. Hold the PS9 at a 30° angle (cup sloping downward).
3. Thoroughly flush the pH sensor well and conductivity cell cup with a steady stream of the solution you intend to measure by allowing the solution to flow into and out of the sensor well and cell cup for at least 10 seconds.
4. Let sample flow continuously into conductivity cell cup with no aeration.


5. Allow both the pH sensor well and conductivity cell cup to remain filled with sample.
6. Press . The instrument will begin alternating between a predicted final ORP value and a free chlorine equivalent concentration in ppm. Both readings will change rapidly at first.
7. Wait for the readings to stabilize. When the mV and ppm values are unchanging for 5 consecutive readings, the FC^E reading has reached a stable level. This may take 1 to 2 minutes.
NOTE: If the reading takes more than 1 minute to stabilize, press the  after 1 minute to prevent the Auto OFF feature from disturbing the measurement process. Annunciators will alert you when either the pH or ORP of the final FC^E are Out of Range ("-Or-").
8. Press **MS** to store reading in memory if desired.

3. FC^E Immersion Method

NOTE: Use this method for pools, spas and other large standing bodies of water.

1. Hold instrument beneath the surface of the water to avoid surface effects on the water's chemistry.
2. Swirl the instrument around for at least 10 seconds to thoroughly rinse the conductivity cell cup and pH sensor well.
3. Continue holding the instrument under the surface while taking the reading.
4. Press .
5. The instrument will begin alternating between a predicted final ORP value and a free chlorine equivalent concentration in ppm. Both readings will change rapidly at first.
6. Wait for the readings to stabilize. When the mV and ppm values are unchanging for 5 consecutive readings, the FC^E reading has reached a stable level. This may take 1 to 2 minutes.

NOTE: If the reading takes longer than 1 minute to stabilize,

press  after 1 minute to prevent the Auto OFF feature from disturbing the measurement process. Annunciators will alert you when either the pH or ORP of the final FC^E ppm value are Out of Range (“-Or-”).

7. Press **MS** to store reading in memory if desired.

4. FC^E Best Practices

For best results it is recommended that you:

1. Take at least 3 consecutive FC^E measurements and record the readings.
2. Discard measurements that are significantly different from the others.
3. Calculate the average of the remaining measurements. Use this value.

IMPORTANT: The sensor well must be kept wet. Before replacing the rubber cap, rinse and fill the sensor well with Myron L[®] pH Sensor Storage Solution.

VI. SOLUTION SELECTION

A. Why Solution Selection is Available

Conductivity, resistivity, and TDS require temperature correction to 25°C values (ref. Standardized to 25°C, pg. 54). Selection determines the temperature correction of conductivity and calculation of TDS from compensated conductivity (ref. Cond. Conversion to TDS, pg. 57).

B. The 3 Solution Types

On the left side of the display is the salt solution characteristic used to model temperature compensation of conductivity and its TDS conversion. Generally, using KCl for conductivity, NaCl for MIN/SALT, and 442TM (Natural Water characteristic) for TDS will reflect present industry practice for standardization. This is how your instrument is shipped from the factory (ref. Solution Characteristics, pg. 57).

C. Calibration of Each Solution Type

There is a separate calibration for each of the 3 solution types. Note that calibration of a 442TM solution does not affect the calibration of a NaCl solution. For example: Calibration (ref. Conductivity, MIN/SALT or TDS Calibration, pg. 27) is performed separately for each type of solution one wishes to measure (ref. Conductivity/TDS Standard Solutions, pg. 52).

D. Procedure to Select a Solution

NOTE: Check display to see if solution displayed (KCl, NaCl or 442™) is already the type desired. If not:

1. Press COND , $\frac{\text{MIN}}{\text{SALT}}$ or TDS to select the parameter on which you wish to change the solution type.
2. Press and hold $\frac{\text{CAL}}{\text{MCLR}}$ key for 3 seconds to make “SEL” appear (see Figure 1). For demonstration purposes, all 3 solution types are shown simultaneously.



Figure 1

3. Use the $\frac{\blacktriangle}{\text{MS}}$ or $\frac{\text{MR}}{\blacktriangledown}$ key to select type of solution desired (ref. Solution Characteristics, pg. 57). The selected solution type will be displayed: KCl, NaCl or 442™.
4. Press $\frac{\text{CAL}}{\text{MCLR}}$ to accept new solution type.

In these first six sections, you have learned all you need to make accurate measurements. The following sections contain calibration, advanced operations and technical information.

VII. CALIBRATION

A. Calibration Intervals

Generally, calibration is recommended about once per month with Conductivity or TDS solutions. Calibration with pH solutions should be checked twice a month. Calibration of ORP is not necessary (ref. CALIBRATION INTERVALS TABLE under Calibration Steps below).

B. Rules for Calibration of the POOLPRO


1. Calibration Steps


The following table lists measurement functions and their corresponding calibration requirements:

CALIBRATION INTERVALS TABLE:

Function	KCl, NaCl or 442™
Cond	Gain Only
MIN/SALT	Gain Only
TDS	Gain Only
pH	7, acid and/or base
ORP	Zero set with pH 7 automatically
Alkalinity	Gain Only
Hardness	Gain Only

a. Starting Calibration


For Alkalinity and Hardness, calibration is begun by pressing  when the titration is complete and a value displays.

For Conductivity, TDS or pH, calibration is begun by pressing 


while measuring Conductivity, TDS or pH. Measuring continues, but the “**CAL**” icon is on, indicating calibration is now changeable. The

reading is changed with the  and  keys to match the known value.


The calibration for each of the 3 solution types may be performed in either conductivity or TDS mode.

Once in CAL mode, the  key becomes an “ACCEPT” key for the new calibration.

For pH, at each point, pressing  accepts the new calibration value and steps you to the next adjustment.

To bypass a calibration, simply press  to accept the present value as-is.

b. Leaving Calibration

Pressing  accepts calibration and exits CAL mode if there are no more steps. Calibration is complete when the “**CAL**” icon goes out. Pressing any measurement key during calibration cancels changes not yet accepted and exits calibration mode. Leaving pH after the 2nd buffer results in the same gain being entered in place of the 3rd buffer.

2. Calibration Limits

There are calibration limits for all functions for which calibration is required. The PS9 will not allow you to calibrate beyond its calibration limits. Attempts to calibrate too far up or down from the ideal “**FAC**” value stored in the PS9 by the factory will cause the displayed value to be replaced with “**FAC**”. If you accept it (press the **CAL** key), you will have the original default factory calibration for this measurement. The need to calibrate so far out that “**FAC**” appears indicates a procedural problem, incorrect standard solution, a very dirty cell cup or an aging pH/ORP sensor (ref. Troubleshooting Chart, pg. 50).

C. Calibration Procedures

1. Conductivity, MIN/SALT or TDS Calibration








- a. Rinse conductivity cell 3 times with proper standard (KCl, NaCl, or 442™) (ref. Cond/TDS Standard Solutions, pg. 52).
- b. Refill conductivity cell with same standard. KCl-7000 shown.
- c. Press ,  or , then press ; “**CAL**” icon will appear on the display (see Figure 2).









Figure 2

- d. Press  or  to step the displayed value toward the standard's value (7032 > 7000) or hold a key down to scroll rapidly through the reading.
- e. Press  once to confirm new value and end the calibration sequence for this particular solution type. If another solution type is also to be measured, change solution type now and repeat this procedure.

2. Reloading Factory Calibration (Cond. MIN/SALT or TDS)

If calibration is suspect or known to be incorrect, and no standard solution is available, the calibration value can be replaced with the original factory value for that solution. This "FAC" value is the same for all POOLPROs, and returns you to a known state without solution in the cell. The "FAC" internal electronics calibration (which bypasses the electrodes and cell) is not intended to replace calibration with conductivity/TDS standard solutions. If another solution type requires resetting, change solution type and repeat this procedure.

- a. Press ,  or .
- b. Press .
- c. Press  key until "**FAC**" appears and release.
- d. Press  to accept the factory calibration setting.

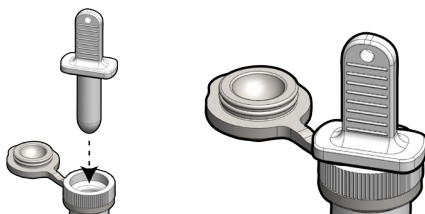
3. Alkalinity Calibration

Prepare the materials required for titration: Alkalinity Standard Solution 100PPM, Reagent: A1, cell extender, T-plunger, and 100 μ L pipette with a clean tip installed.

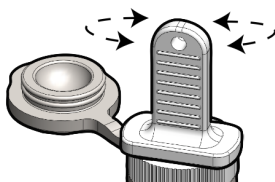
NOTE: If you do not complete each titration point within the 3-minute timeout period, the PS9 exits the Alkalinity function. Start the titration process over.

If you make a gross error in your titration method, e.g., you forget to add sample to the cell, “**Err**” will display. Start the titration over.

1. Ensure pH protective cap is secure to avoid contamination.
2. Install the cell extender by pushing the base of the cell extender into the conductivity cell until it is fully seated.
3. Rinse the conductivity cell cup and cell extender 3 times with Alkalinity Standard 100PPM.
4. Refill cell cup and cell extender with Alkalinity Standard 100PPM.
5. Insert T-plunger into open cell extender until the “T” rests flush on the lip of the cell extender. This will cause fluid beyond what is required for titration to overflow the cell extender.



6. Rotate the T-plunger from side to side, allowing the arms of the T-plunger to slide along the rim of the cell extender.



7. Carefully remove the T-plunger so as not to flick or spill remaining sample. Tap the T-plunger on the inside rim of the cell extender to remove sample that may be stuck to the T-plunger.
8. Press **ALK**. “**AL**” alternating with “**tdS**” briefly displays along with the “**PPM**” value of the sample. “**PRES CAL**” will display when the reading stabilizes.

9. Press **CAL** to accept and advance to the first titration prompt. "Add" alternating with "A1" displays
10. Using the pipette, add 100 μ L of Reagent: A1, close the cell extender cap securely, press **CAL** to continue.
11. "AGit t1" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue. (The number after "t" indicates which titration point you are on. The number "1" here indicates this is the first titration point. Subsequent titration points are sequenced numerically: 2, 3, 4, etc.).
12. "HOLD" displays. Keep holding the meter steady until "Add" alternating with "A1" displays. Open the cell extender cap carefully to avoid spilling, and use the pipette to add 100 μ L of Reagent A1. Close the cell extender cap securely. Press **CAL** to continue.
13. Repeat Steps 11 and 12 until the meter reports the alkalinity result.
14. Press **CAL**. The "CAL" icon will display.
15. Adjust the displayed value to "100" using the **UP** or **DOWN** key.

NOTE: If you scroll beyond $\pm 10\%$ of the reading "FAC" will display. If you press the **CAL** key while "FAC" is displayed, the instrument will return Alkalinity to factory calibration (erasing any previous wet calibration adjustments).

16. Press **CAL** to accept. The PS9 will automatically power off.

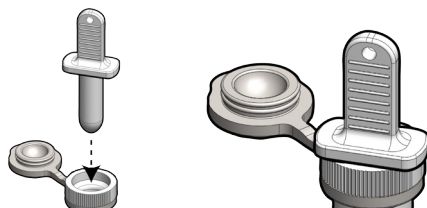
4. Hardness Calibration

Prepare the materials required for titration: Hardness Standard Solution 200PPM, Reagent: HB, Reagent: H4, or Reagent: H5, cell extender, T-plunger, 100 μ L pipette with a clean tip installed, and one extra clean tip.

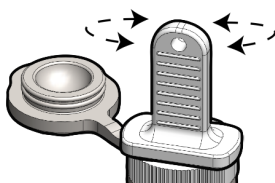
NOTE: If you do not complete each titration point within the 3-minute timeout period, the PS9 exits the Hardness function. Start the titration process over.

1. Ensure pH protective cap is secure to avoid contamination.
2. Install the cell extender by pushing the base of the cell extender into the conductivity cell until it is fully seated.
3. Rinse the conductivity cell cup and cell extender 3 times with Hardness Standard 200PPM.

4. Refill cell cup and cell extender with Hardness Standard 200PPM.
5. Insert T-plunger into open cell extender until the "T" rests flush on the lip of the cell extender. This will cause fluid beyond what is required for titration to overflow the cell extender.



6. Rotate the T-plunger from side to side, allowing the arms of the T-plunger to slide along the rim of the cell extender.



7. Carefully remove the T-plunger so as not to flick or spill remaining sample. Tap the T-plunger on the inside rim of the cell extender to remove sample that may be stuck to the T-plunger.

8. Press **HARD**. "**CA**" alternating with "**tdS**" displays along with "**PPM**" sample value. "**PRES CAL**" displays when the reading stabilizes.

9. Press **CAL**. "**HC**" or "**LC**" will display; use the **UP** or **DOWN** keys to switch to **HC** for the higher range (0-1710 ppm) or **LC** for the lower range (0-200ppm). Press **CAL** to accept and continue.

10. "**Add**" alternating with "**buFF**" displays. Using the pipette, add 100 μ L of Reagent: HB to the sample, close the cell extender cap securely. Change the pipette tip for adding **EDTA** in following steps. Press **CAL** to continue.

11. "**AGit**" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue.

12. "**HOLD**" displays. Keep holding the meter steady until "**Add**" alternating with "**EdtA**" displays. Open the cell extender cap carefully to avoid spilling, and use the pipette to add 100 μ L of **EDTA-B-HC** or **EDTA-B-LC**, depending on the previous selection. Close the cell extender cap securely. Press **CAL** to continue.

13. "AGit t1" flashes on the screen. Agitate to mix the sample. When agitation is done, tap the cell extender cap to dispel any solution clinging to the cap back into the cell. Press **CAL** to continue. (The number after "t" indicates which titration point you are on. The number "1" here indicates this is the first titration point. Subsequent titration points are sequenced numerically: 2, 3, 4, etc.)
14. Repeat Steps 12 and 13 until the meter reports the hardness result.
15. Press **CAL**. The "CAL" icon will display.
16. Adjust the displayed value to "200" using the **UP** or **DOWN** key.



Note: If you scroll beyond $\pm 10\%$ of the reading "FAC" will display. If you press the **CAL** key while "FAC" is displayed, the instrument will return Hardness to factory calibration (erasing any previous wet calibration adjustments).

17. Press **CAL** to accept. The PS9 will automatically power off.

5. pH Calibration

IMPORTANT: Always "zero" your POOLPRO with a pH 7 buffer solution before adjusting the gain with acid or base buffers, i.e., 4 and/or 10, etc.

a. pH Zero Calibration

1. Rinse sensor well and cell cup 3 times with 7 buffer solution.
2. Refill both sensor well and cell cup with 7 buffer solution.
3. Press  to verify the pH calibration. If the display shows 7.00, skip the pH Zero Calibration and proceed to section b. pH Gain Calibration.
4. Press  to enter calibration mode. The "CAL", "BUFFER" and "7" annunciators will appear (see Figure 3). Displayed value will be the uncalibrated sensor.

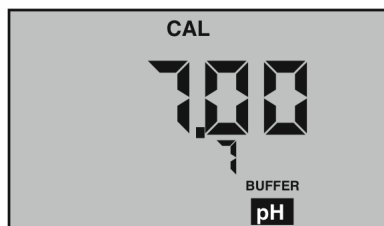






Figure 3

NOTE: If a wrong buffer is added (outside of 6-8 pH), “7” and “**BUFFER**” will flash, and the POOLPRO will not adjust. The uncalibrated pH value displayed in Step 4 will assist in determining the accuracy of the pH sensor. If the pH reading is above 8 with pH 7 buffer solution, the sensor well needs additional rinsing or the pH sensor is defective and needs to be replaced.

5. Press  or  until the display reads “7.00”.

NOTE: Attempted calibration of >1 pH point from factory calibration will cause “**FAC**” to appear. This indicates the need for sensor replacement (ref. Troubleshooting pg. 50) or fresh buffer solution. The “FAC” internal electronic calibration is not intended to replace calibration with pH buffers. It assumes an ideal pH sensor Each “FAC” indicates a factory setting for that calibration step (i.e., 7, acid, base).


You may press  to accept the preset factory value, or you may reduce your variation from factory setting by pressing  or .

6. Press  to accept the new value. The pH Zero Calibration is now complete. You may continue with pH Gain Calibration or exit by pressing any measurement key.

b. pH Gain Calibration

IMPORTANT: Always calibrate or verify your POOLPRO with a pH 7 buffer solution before adjusting the gain with acid or base buffers, i.e., 4 and/or 10, etc. Either acid or base solution can be used for the 2nd point “Gain” calibration and then the opposite for the 3rd point. The display will verify that a buffer is in the sensor well by displaying either “**Acid**” or “**bAS**”.

1. The pH calibration mode is initiated by either completion of the pH Zero Calibration, or verifying 7 buffer and pressing the

 key twice while in pH measurement mode.

2. At this point the “**CAL**”, “**BUFFER**” and “**Acid**” or “**bAS**” annunciators will be displayed (see Figures 4 and 5, pg. 34).

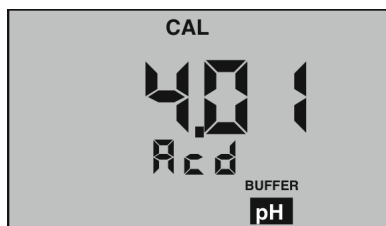


Figure 4

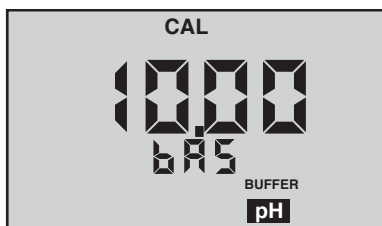






Figure 5

NOTE: If the “Acid” and “bAS” indicators are blinking, the PS9 is indicating an error and needs either an acid or base solution present in the sensor well.

3. Rinse sensor well 3 times with acid or base buffer solution.
4. Refill sensor well again with same buffer solution.
5. Press  or  until display agrees with buffer value.
6. Press  to accept 2nd point of calibration. Now the display indicates the next type of buffer to be used.

Single point Gain Calibration is complete. You may continue for the 3rd point of calibration (2nd Gain) or exit by pressing any measurement key. Exiting causes the value accepted for the buffer to be used for both acid and base measurements.

To continue with 3rd point calibration, use basic buffer if acidic buffer was used in the 2nd point, or vice-versa. Again, match the display to the known buffer value as in Step 2 and continue with the following steps:

7. Repeat Steps 3 through 6 using opposite buffer solution.
8. Press  to accept 3rd point of calibration, which completes the calibration procedure. Fill sensor well with Myron L® Storage Solution and replace protective cap.

6. ORP Calibration

ORP electrodes rarely give false readings without problems in the reference electrode. For this reason, and because calibration solutions for ORP are highly reactive and potentially hazardous, your POOLPRO has an electronic ORP calibration. This causes the zero point on the reference electrode to be set whenever pH 7 calibration is done.

7. Temperature Calibration

Temperature calibration is not necessary in the POOLPRO.

VIII. CALIBRATION INTERVALS

There is no simple answer as to how often one should calibrate an instrument. The POOLPRO is designed to not require frequent recalibration. The most common sources of error were eliminated in the design, and there are no mechanical adjustments. Still, to ensure specified accuracy, any instrument must be checked against chemical standards occasionally.

A. Suggested Intervals

On the average, we expect calibration need only be checked monthly for the Conductivity, MIN/SALT or TDS functions. The pH function should be checked every 2 weeks to ensure accuracy. Measuring some solutions will require more frequent intervals.

B. Calibration Tracking Records

To minimize your calibration effort, keep records. If adjustments you are making are minimal for your application, you can check less often. Changes in conductivity calibration should be recorded in percent. Changes in pH calibration are best recorded in pH units.

Calibration is purposely limited in the POOLPRO to $\pm 10\%$ for the conductivity cell, as any change beyond that indicates damage, not drift. Likewise, calibration changes are limited to ± 1 pH unit, as any change beyond that indicates the end of the sensor's lifetime and replacement is recommended.

C. Conductivity, MIN/SALT & TDS, Practices to Maintain Calibration

1. Clean oily films or organic material from the cell electrodes with foaming cleaner or mild acid. Do not scrub inside the cell.
2. Calibrate with solutions close to the measurements you make. Readings are compensated for temperature based on the type of solution. If you choose to measure tap water with a KCl compensation, which is often done (ref. An Example of 2 different solution selections and the resulting compensation, pg. 55), and you calibrate with 442™ solution because it is handy, the further away from 25°C you are, the more error you have. Your records of calibration changes will reflect temperature changes more than the instrument's accuracy.

3. Rinse out the cell with pure water after taking measurements. Allowing slow dissolving crystals to form in the cell contaminates future samples.
4. For maximum accuracy, keep the pH sensor cap on tight so that no fluid washes into the conductivity cell.

D. pH and ORP/Free Chlorine Equivalent Practices to Maintain Calibration

1. The sensor well must be kept wet. Before replacing the rubber cap, rinse and fill the sensor well with Myron L[®] pH Sensor Storage Solution. Although not ideal, if Myron L[®] Storage Solution is not available, you can temporarily use a strong KCl solution, a pH 4 Buffer Solution, or a saturated solution of table salt and tap water until pH Storage Solution is obtained. NEVER use distilled water.
2. Rinse away caustic solutions immediately after use.

ORP calibration solutions are caustic, and $\pm 5\%$ is considered very accurate. By using the pH zero setting (0 mV = 7 pH) for ORP and precision electronics for detection, the POOLPRO delivers better accuracy without calibration than a simpler instrument could using calibration solutions.

IX. MEMORY

This feature allows up to 100 readings with their temperatures to be stored simultaneously for later recall. At the same time, the TIME and DATE are also recorded. To download the memory to a computer, ref. bluDock™ WIRELESS DATA TRANSFER INSTRUCTIONS, pg. 45.

A. Memory Storage






1. While displaying a measurement, press  to record the displayed value.
2. “**MEMORY**” will appear and the temperature display will be momentarily replaced by a number (1-100) showing the position of the record. Figure 6 shows a reading of 1806 μS stored in memory record #4.




Figure 6

B. Memory Recall






1. Press any measurement key.
2. Press ; “**MEMORY**” will appear, and the display will show the last record stored.
3. Press  or  to scroll to the record location desired (the temperature display alternates between temperature recorded and location number).
4. Press  to display time and date stamp.
5. Press any measurement key to leave memory recall or allow to automatically turn off.

C. Clearing a Record/Memory Clear

After recalling a certain record location, press and HOLD  to

clear that memory. This space will be the place for the next memory record, unless you scroll to another empty position before ending the recall sequence. The next memory stored will go into the next highest available memory location.

Example: You have locations 1-7 filled and wish to clear the conductivity reading stored in record location #3 and replace it with a pH reading.

1. Press  and scroll to location #3.
2. Press and HOLD  to clear old record #3.
3. Fill pH/ORP sensor well with sample.
4. Press  to measure sample and press  to store reading in location #3.
5. The next memory stored will go into location #8.
6. To clear **all** records: After pressing , scroll down. “**CLR ALL**” will be displayed (see Figure 7, pg. 38).

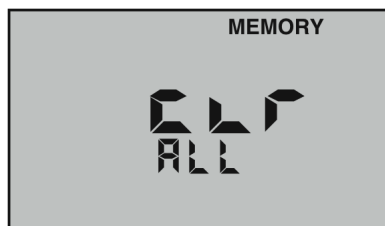


Figure 7

7. Press $\left(\frac{\text{CAL}}{\text{MCLR}} \right)$. All records will be cleared.

X. TIME and DATE

The Time and Date may easily be changed as you travel.

A. Setting TIME

Time is always displayed in 24 hour time.

Example shown in Figure 8, 16:05 equals 4:05 PM.

1. Press $\left(\text{COND} \right)$.
2. Press $\left(\frac{\text{MR}}{\blacktriangledown} \right)$ until the time is displayed (scrolling through stored readings, “**PC OFF**”, and “**CLr ALL**” to time, e.g., “**16:05**”).
3. Press $\left(\frac{\text{CAL}}{\text{MCLR}} \right)$ to initiate. “**CAL**” will be displayed along with the time (see Figure 8).



Figure 8

4. Press $\left(\frac{\blacktriangle}{\text{MS}} \right)$ or $\left(\frac{\text{MR}}{\blacktriangledown} \right)$ to change the time.
5. Press $\left(\frac{\text{CAL}}{\text{MCLR}} \right)$ to accept the change (new time).

B. Setting DATE

Example shown in Figure 9 is in US format, i.e., mo/dy/yr.

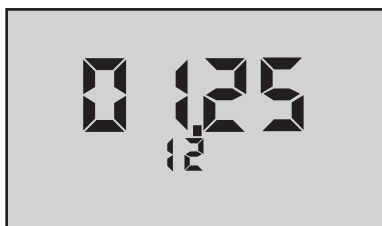


Figure 9

NOTE: The default format is US. Date format may be changed (ref. Date Format “US and International (Int)”, pg. 40).










1. Press .
2. Press  repeatedly until the date is displayed (scrolling through stored readings, “PC OFF”, “CLF ALL” and time to the date, e.g., “01.25 12” (Figure 9), January 25, 2012).
3. Press  to initiate. “CAL” will be displayed along with the YEAR (see Figure 10).



Figure 10

4. Press  or  to change the YEAR.
5. Press  to accept the change (new year).
6. Press  or  to change the month.
7. Press  to accept the change (new month), (see Figure 11, pg. 40).

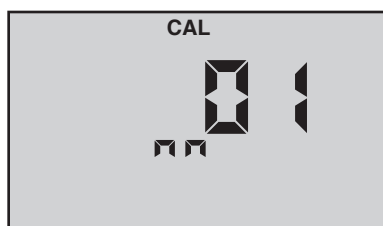


Figure 11

8. Press the $\frac{\blacktriangle}{\text{MS}}$ or $\frac{\text{MR}}{\blacktriangledown}$ to change the day.
9. Press $\frac{\text{CAL}}{\text{MCLR}}$ to accept the change (new day) (see Figure 12).

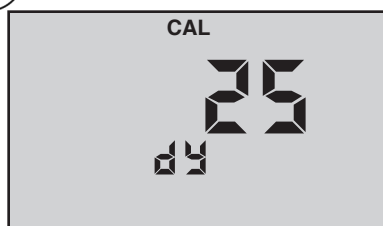


Figure 12

C. DATE FORMAT "US & International (Int)"

1. Press $\frac{\text{COND}}{\text{MCLR}}$.
2. Press $\frac{\text{MR}}{\blacktriangledown}$ repeatedly until the format is displayed (scrolling through stored readings, "**PC OFF**", "**CLR ALL**", time and date to date **format**).
3. Press $\frac{\text{CAL}}{\text{MCLR}}$ to change. Display will now indicate other format (see Figures 13 & 14, pg.41).

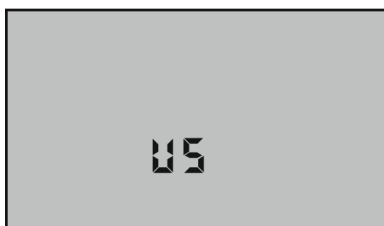


Figure 13

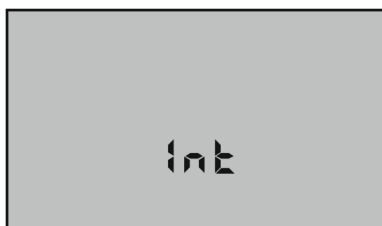





Figure 14

4. Press any measurement key or allow to automatically turn off.

XI. TEMPERATURE FORMAT “Centigrade & Fahrenheit”

1. Press .
2. Press  to display the stored memory records.
3. Press  repeatedly until you pass the “US” or “Int” date format location. The display will show a “C” or “F” (see Figures 15 and 16).

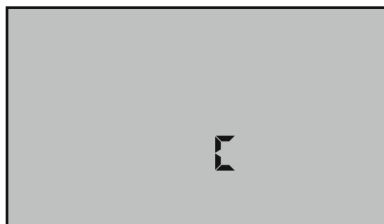


Figure 15

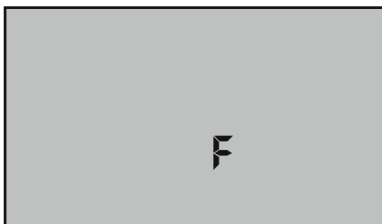




Figure 16

4. Press  to switch units.
5. Press  to accept unit preference for all temperature readings.

NOTE: Tempco will still be shown in %/°C.

XII. TOTAL RETURN to FACTORY SETTINGS “FAC SEL”

There may come a time when it would be desirable to quickly reset all the recorded calibration values in the instrument back to the factory settings. This might be to ensure all calibrations are set to a known value, or to give the instrument to someone else free of adjustments or recorded data for a particular application.

NOTE: All stored data will be lost.





1. Press .
2. Press  to display the stored memory records.
3. Press  repeatedly until you pass the “**CLR ALL**” and the C-F locations. The display will show a “**FAC SEL**” (see Figure 17).



Figure 17

4. Press  to accept the resetting. Display will return to Conductivity.

XIII. CELL CHECK

The cell check verifies the cleanliness of the conductivity/MIN/SALT/TDS sensor. In normal use the cell may become dirty or coated and require cleaning. If the display is showing “**.00**” when the cell cup is dry, the sensor is probably clean. No matter what a manufacturer claims, a sensor can and will become contaminated or coated and, therefore, require cleaning. A true 4-wire sensor, as in the POOLPRO, helps to mitigate contamination, but **NO SENSOR IS 100% IMMUNE.**





1. Press .
2. Press  to display the stored memory records.
3. Press  repeatedly until you pass the FAC SEL location. The display will show a “**CELL ch**” (see Figure 18, pg. 43).



Figure 18

4. Press  to test. If cell is clean, “Good” will momentarily be displayed (see Figure 19). If cell is dirty, “CELL cLn” will be displayed (see Figure 20) (ref. Cleaning Sensors, pg. 47).

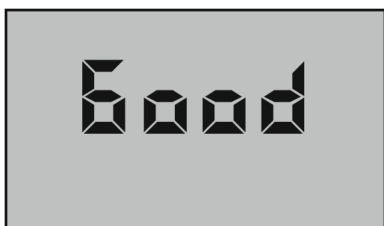


Figure 19



Figure 20

XIV. **AUTO OFF**

Auto off allows the user to adjust the time the instrument is ON (up to 75 seconds) after each press of a key. Default time is 15 seconds with 60 seconds in “**CAL**” (calibration) mode.





1. Press .
 2. Press  to display the stored memory records.
 3. Press  repeatedly until you pass the “**CELL ch**” location.
- The display will show “**Auto oFF**” (see Figure 21).



Figure 21

4. Press  to initiate. "CAL" will be displayed along with "15 SEC" or current Auto oFF value (see Figure 22).

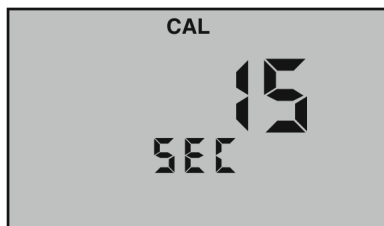




Figure 22

5. Press  or  to change the amount of time (see Figure 23). Maximum time of 75 seconds is shown.

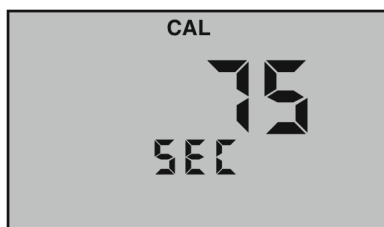



Figure 23

6. Press  to accept the change (new time).

XV. **bluDock™ WIRELESS DATA TRANSFER INSTRUCTIONS**

NOTE: The Bluetooth® word mark and logos are registered trademarks owned by the Bluetooth® SIG, Inc. and any use of such marks by Myron L® Company is under license.

Requires a bluDock enabled instrument. If your PS9 does not include the bluDock option ("-BD") you may purchase a Myron L® bluDock accessory package, Model # BLUDOCK. Package includes POOLPRO hardware modification that allows the PS9 to communicate wirelessly with a personal computer configured for wireless device communication. Package includes a bluDock software, Guardian²™, that will operate on Windows®* 7 or later/macOS®** 10.13 or later based computer systems.

* Windows® is a registered trademark of Microsoft Corporation.

** macOS® is a registered trademark of Apple, Inc

A. Software Installation


1. Download Guardian²™ application from the Myron L[®] website: <http://www.myronl.com/>. Select the Downloads tab and click on Guardian² Applications.
2. Download and install the Guardian² application for your operating system. You will also find the Quick Start Guide and Operation Manual for the Guardian² Application on this webpage.
Additional drivers may be required. See our website for the latest information.

B. Hardware Setup

For a computer without Bluetooth[®] capability: A Bluetooth dongle may be purchased from your local electronics store. Plug in the dongle and install per manufacturer's instructions.

For computers with Bluetooth capability/Bluetooth dongle installed:

First time use of the bluDock:

1. Press any parameter button to turn the POOLPRO on.
2. Put the POOLPRO in "PC On" mode by pressing the  key until "PC OFF" appears (see Figure 24).

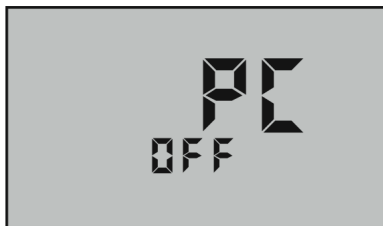



Figure 24

3. Then press the  key. "PC On" will be displayed (see Figure 25).

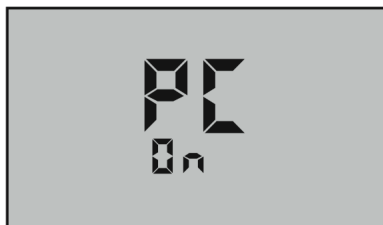


Figure 25

NOTE: “PC Ini” may momentarily be displayed while initializing (see Figure 26).

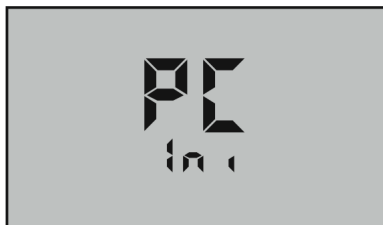


Figure 26

4. The bluDock device will appear as MyronL-XXXX in your computer's Bluetooth devices list. While pairing with the bluDock device it will ask for the pin, **THE BLUDOCK DEVICE PIN IS 1234.**
5. After pairing, note the bluDock number for your device, MyronL-XXXX.
(For complete instructions see the Guardian² Operation Manual for your computer operating system.)

NOTE: The PS9 will automatically power down after 3 minutes. If the PS9 powers down during pairing, repeat Steps 1-3 above and continue.

C. Memory Stack Download

1. With the POOLPRO in “PC On” mode, open the software application.
2. Verify that the device name selected matches the bluDock number for your device. (MyronL-XXXX)
3. In the application, click on the data download button. A data transfer bar will appear while the data is being downloaded.

Once downloaded, the data may be manipulated, or printed within the Myron L[®] application, or the data may be exported to spreadsheet formats such as Microsoft Excel[®] (.xls and .xlsx) or universal formatting (.csv). The user can also save data in a secure, encrypted format (.mlcx).

Please Note: Although the Myron L[®] Company has performed extensive testing, we cannot guarantee compatibility of all applications and formats. We suggest testing your application and format for compatibility before relying on it.

Additional features such as assigning a name to the PS9, setting time and date and erasing data are available. Visit our website for the latest instructions: <http://www.myronl.com>

4. Turn off POOLPRO PC download mode by selecting any measurement function. Failure to do so will reduce battery life.

XVI. CARE and MAINTENANCE

- POOLPROs should be rinsed with clean water after use.
- Solvents should be avoided.
- Do not drop, throw or otherwise strike the instrument. Shock damage may cause instrument failure.

A. Temperature Extremes

Solutions in excess of 71°C/160°F should not be placed in the cell cup area; this may cause damage. The pH sensor may fracture if the POOLPRO temperature is allowed to go below 0°C/32°F. Care should be exercised not to exceed rated operating temperature.

Leaving the POOLPRO in a vehicle or storage shed on a hot day can easily subject the instrument to over 66°C/150°F. This will void the warranty.

B. Battery Replacement

Dry Instrument THOROUGHLY. Remove the four (4) bottom screws. Open instrument carefully. Carefully detach battery from circuit board. Replace with 9 Volt alkaline battery. Replace bottom, ensuring the sealing gasket is installed in the groove of the top half of case. Re-install screws, tighten evenly and securely.

NOTE: Because of nonvolatile EEPROM circuitry, all data stored in memory and all calibration settings are protected even during power loss or battery replacement. However, loss of time and date may occur if battery is removed for longer than 3 minutes (180 seconds).

C. pH/ORP Sensor Replacement

Order model RPR. When ordering, be sure to include the model and serial number of your instrument to ensure receipt of the proper type. Complete installation instructions are provided with each replacement sensor.

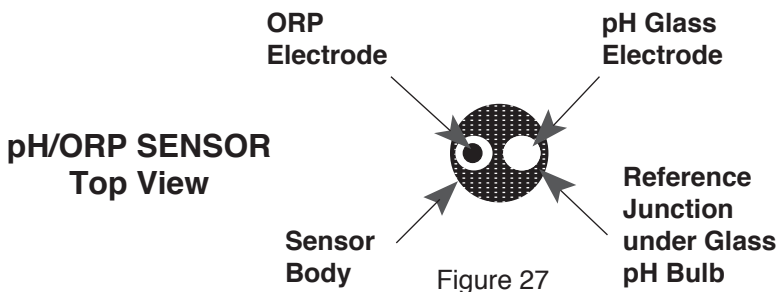
D. Cleaning Sensors

1. Conductivity/MIN/SALT/TDS

The conductivity cell cup should be kept as clean as possible. Flushing with clean water following use will prevent buildup on electrodes. However, if very dirty samples — particularly scaling types — are allowed to dry in the cell cup, a film will form. This film reduces accuracy. When there are visible films of oil, dirt, or scale in the cell cup or on the electrodes, use isopropyl alcohol or a foaming non-abrasive household cleaner. Rinse out the cleaner and your POOLPRO is again ready for accurate measurements.

2. pH/ORP

The unique pH/ORP sensor in your POOLPRO is a nonrefillable combination type that features a porous liquid junction. (See Figure 27.) It should not be allowed to dry out. However, if this occurs, the sensor may sometimes be rejuvenated by first cleaning the sensor well with Isopropyl alcohol or a liquid spray cleaner such as Windex™ or Fantastic™ and rinsing well. Do not scrub or wipe the pH/ORP sensor.



Then use one of the following methods:

1. Pour a HOT salt solution ~60°C/140°F — a potassium chloride (KCl) solution such as Myron L® pH/ORP Sensor Storage Solution is preferable, but HOT tap water with table salt (NaCl) will work fine — in the sensor well and allow to cool. Retest.

or

2. Pour DI water in the sensor well and allow to stand for no more than 4 hours (longer can deplete the reference solution and damage the glass bulb). Retest.

If neither method is successful, the sensor must be replaced.

“Drifting” can be caused by a film on the pH sensor bulb and/or reference. Use isopropyl alcohol (IPA) or spray a liquid cleaner such as Windex™ or Fantastic™ into the sensor well to clean it. The sensor bulb is very thin and delicate. Do not scrub or wipe the pH/ORP sensor.

Leaving high pH (alkaline) solutions in contact with the pH sensor for long periods of time is harmful and will cause damage. Rinse such liquids from the pH/ORP sensor well and refill it with Myron L® Storage Solution to extend the useful life of the sensor.

Samples containing chlorine, sulfur, or ammonia can “poison” any pH electrode. If it is necessary to measure the pH of any such sample, thoroughly rinse the sensor well with clean water immediately after taking the measurement. Any sample element that reduces (adds an electron to) silver, such as cyanide, will attack the reference electrode.

Replacement sensors are available only from the Myron L® Company or its authorized distributors (ref. Replacement pH/ORP Sensor, pg. 54).

It is possible to desensitize the ORP sensor when measuring high ORP solutions. To check the ORP sensor function, take an ORP reading of Myron L® pH/ORP sensor storage solution. If the reading is outside the range of 350-400 mV, clean ONLY the platinum ORP electrode with Myron L® ORP Conditioner solution-soaked cotton swab, being careful not to touch the swab to the glass bulb of the pH sensor.

XVII. TROUBLESHOOTING CHART

Symptom	Possible Cause
No display , even though measurement key pressed	Battery weak or not connected.
Inaccurate pH readings	<ol style="list-style-type: none"> 1. pH calibration needed. Ref. pH Cal., pg. 33. 2. Cross-contamination from residual pH buffers or samples in sensor well. 3. Calibration with expired pH buffers.
No response to pH changes	Sensor bulb is cracked or an electromechanical short caused by an internal crack.
Will not adjust down to pH 7	pH/ORP sensor has lost KCl.
pH readings drift or respond slowly to changes in buffers/samples or " FAC " is displayed repeatedly	<ol style="list-style-type: none"> 1. Temporary condition due to memory" of solution in pH sensor well for long periods. 2. Bulb dirty or dried out. 3. Reference junction clogged or coated.
Unstable Conductivity/TDS/ MIN/SALT readings	Dirty electrodes.
Unable to calibrate Conductivity/TDS/ MIN/SALT	Film or deposits on electrodes.
Alkalinity or Hardness titration with calibration solution value not as stated on bottle	<ol style="list-style-type: none"> 1. Operation error. 2. Dirty Cell Cup. 3. Dirty pipette. 4. Faulty pipette. 5. Contaminated calibration solution.
Low ORP Reading Slow or no response to ORP changes	ORP platinum electrode is dirty.
FC ^E responds very slowly or returns an atypically high Predictive ORP value	<ol style="list-style-type: none"> 1. Dirty platinum electrode (see above). 2. ORP sensor memory/battery effect. Some ORP sensors exhibit a residual charge when measuring LOW Free Chlorine concentrations soon after measuring a HIGH Free Chlorine concentration.

	Corrective Action
	Check connections or replace battery. Ref. Battery Replacement, pg. 47.
	<ol style="list-style-type: none"> 1. Recalibrate instrument. 2. Thoroughly rinse sensor well. 3. Recalibrate using fresh buffers. Ref. pH Buffer Solutions, pg. 53.
	Replace pH/ORP sensor. Ref. Replacement pH/ORP Sensor, pg. 54.
	Clean and rejuvenate sensor (ref. Cleaning Sensors, pg. 47) and recalibrate. If no improvement, replace pH/ORP sensor (ref. Replacement pH/ORP Sensor, pg. 54).
	Clean and rejuvenate sensor (ref. Cleaning Sensors, pg. 47) and recalibrate. If no improvement, replace pH/ORP sensor (ref. Replacement pH/ORP Sensor, pg. 54).
	Clean cell cup and electrodes. Ref. Cleaning Sensors, pg. 47.
	Clean cell cup and electrodes. Ref. Cleaning Sensors, pg. 47.
	<ol style="list-style-type: none"> 1. Repeat titration. 2. Clean cell cup and electrodes. Ref. Cleaning Sensors, pg. 47. 3. Clean pipette per manufacturer's instructions. 4. Use a new pipette. 5. Use fresh solutions. Be sure to use a new tip for each type of reagent used.
	Check the ORP sensor function. Take an ORP reading of Myron L [®] pH/ORP Sensor Storage Solution (ref. pH Sensor Storage Solution, pg. 53). If the reading is outside the range of 350-400 mV, clean ONLY the platinum ORP electrode with Myron L [®] ORP Conditioner solution-soaked cotton swab (ref. ORP Sensor Conditioner Solution, pg. 53), being careful not to touch the swab to the glass bulb of the pH sensor.
	<ol style="list-style-type: none"> 1. Rinse the pH/ORP sensor well briefly with a small amount of ORP Sensor Conditioner Solution. DO NOT leave the conditioning solution in the sensor well for more than 10 seconds. 2. Rinse the pH/ORP sensor 3 times with Sensor Storage Solution. 3. Fill the sensor well with Sensor Storage Solution and let rest for 5 minutes.

XVIII. ACCESSORIES

NOTE: MSDSs are available on the Myron L® website for all solutions:
http://www.myronl.com/main/Material_Safety_DS_DL.htm

A. Conductivity/TDS Standard Solutions

Your POOLPRO has been factory calibrated with the appropriate Myron L® Company NIST traceable KCl, NaCl, and our own 442™ standard solutions. Most Myron L® conductivity standard solution bottles show three values referenced at 25°C: Conductivity in microsiemens/micromhos, the ppm/TDS equivalents (based on our 442 Natural Water™) and NaCl standards. All standards are within ±1.0% of reference solutions. Available in 2 oz., quarts/liters, and gallon/~3.8 liter bottles.

1. Potassium Chloride (KCl)

The concentrations of these reference solutions are calculated from data in the International Critical Tables, Vol. 6. The 7000 μS is the recommended standard. Order KCL-7000.

2. 442 Natural Water™

442 Natural Water Standard Solutions are based on the following salt proportions: 40% sodium sulfate, 40% sodium bicarbonate, and 20% sodium chloride, which represent the three predominant components (anions) in freshwater. This salt ratio has conductivity characteristics approximating fresh natural waters and was developed by the Myron L® Company over five decades ago. It is used around the world for measuring both conductivity and TDS in drinking water, ground water, lakes, streams, etc. 3000 ppm is the recommended standard. Order 442-3000.

3. Sodium Chloride (NaCl)

This is especially useful in seawater mix applications, as sodium chloride is the major salt component. Most Myron L® standard solution labels show the ppm NaCl equivalent to the conductivity and to ppm 442™ values. The 7500 ppm is the recommended standard. Order NACL-7500.

B. Titration Standard Solutions

Myron L® titration standard solution bottles show both the actual chemical in solution and the ppm calcium carbonate (CaCO_3) equivalent. All standards are within ±1.0% of reference solutions. Available in 2 oz., and quart/liter bottles.

1. Alkalinity Standard Solution

ALK(NaHCO_3)-100 is a solution of sodium bicarbonate (NaHCO_3) equivalent to 100 ppm calcium carbonate that is used for the alkalinity calibration: Order ALK-100.

2. Hardness Standard Solution

HARD(CaCl₂)-200 solution is a solution of calcium chloride (CaCl₂) equivalent to 200 ppm calcium carbonate that is used for the hardness calibration: Order HARD-200.

C. Titration Reagent Solutions

Titration reagents are required for all conductometric titrations - alkalinity and hardness. Reagents are available in 2 oz. and quarts/liters.

1. Reagent: A1

Reagent: A1 is a $\leq 1\%$ WV solution of sulfuric acid (H₂SO₄-1) required for alkalinity titration: Order H2SO4-1.

2. Reagent: HB

Reagent: HB is a $\leq 15\%$ WV solution of pH buffer required for hardness titration: Order HBUFF.

3. Reagent: H4

Reagent: H4 is a low concentration $\leq 5\%$ WV solution of Ethylenediaminetetraacetic acid (EDTA) required for hardness titration: Order EDTA-B-LC.

4. Reagent: H5

Reagent: H5 is a high concentration $\leq 10\%$ WV solution of Ethylenediaminetetraacetic acid (EDTA) required for hardness titration: Order EDTA-B-HC.

D. pH Buffer Solutions

pH buffers are available in pH values of 4, 7 and 10. Myron L® Company buffer solutions are traceable to NIST certified pH references and are color-coded for instant identification. They are also mold inhibited and accurate to within ± 0.01 pH units @ 25°C. Order 4, 7 or 10 Buffer. Available in 2 oz., quarts/liters, and gallon/~3.8 liter bottles.

E. pH Sensor Storage Solution

Myron L® pH Sensor Storage Solution prolongs the life of the pH sensor. Available in 2 oz., quarts/liters, and gallon/~3.8 liter bottles. Order SS.

F. ORP Sensor Conditioner Solution

Myron L® ORP Conditioner Solution removes contaminants and conditions the ORP electrode. Available in 1 oz. Order ORPCOND1OZ.

G. Soft Protective Carry Cases

Padded Nylon carrying case features a belt clip for hands-free mobility. Two colors to choose from:

Blue - Model #: UCC

Desert Tan - Model #: UCCDT

H. Hard Protective Carry Cases

Large, foam-lined case without PS9 includes the 100 μ L pipette, 12 disposable pipette tips, 3 buffers (pH 4, 7, and 10), pH/ORP sensor storage solution, ORP Conditioner solution, reagents A1, HB, H4, H5, and standard solutions, KCl-7000, 442-3000, NaCl-7500, ALK-100 and HARD-200. All bottles are 2 oz/59 ml (except for ORP Conditioner solution which is 1 oz.). Model: TKPS9B.

Small case (for instrument only, no solutions) - Model #: UPP

I. Replacement pH/ORP Sensor

pH/ORP sensor is gel filled and features a unique porous liquid junction. It is user-replaceable and comes with easy to follow instructions.

Model #: RPR

J. Pipette Kit with Replacement Tips

3 color coded M'Pet pipettes with 48 replacement tips. This allows you to have a dedicated pipette for each titration reagent. Colored dots on each pipette match the colored dots on reagent solution bottles to help you avoid confusion and contamination. Model #: PROKITB

K. Replacement M'Pet Pipette

1 replacement 100 μ L fixed volume pipette. Model #: FVMP-100

L. M'Pet Pipette Replacement Tips

Bag of 48 pipette tips. Model #: PTIP100-48

M. bluDock™ Wireless Data Transfer Accessory Package

This accessory allows the operator to download the POOLPRO memory stack to a spreadsheet on a computer. The package includes bluDock modified circuit board to be installed inside the PS9, software installation and operating instructions. Model #: BLUDOCK

XIX. TEMPERATURE COMPENSATION (Tempco) of Aqueous Solutions

Electrical conductivity indicates solution concentration and ionization of the dissolved material. Since temperature greatly affects ionization, conductivity measurements are temperature dependent and are normally corrected to read what they would be at 25°C.

A. Standardized to 25°C

Conductivity is measured with great accuracy in the POOLPRO using a method that ignores fill level, electrolysis, electrode characteristics, etc., and features a microprocessor to perform temperature compensation. In simpler instruments, conductivity values are usually assigned an average correction similar to that of KCl solutions for correction to 25°C. The correction to an equivalent KCl solution is a standard set by chemists that standardizes the measurements and allows calibration with precise KCl solutions. In the POOLPRO, this correction can be set to other solutions or tailored for special measurements or applications.

B. Tempco Variation

Most conductivity instruments use an approximation of the temperature characteristics of solutions, perhaps even assuming a constant value. The value for KCl is often quoted simply as 2%/°C. In fact, KCl tempco varies with concentration and temperature in a non-linear fashion. Other solutions have more variation still. The POOLPRO uses corrections that change with concentration and temperature instead of single average values. See Chart 1.

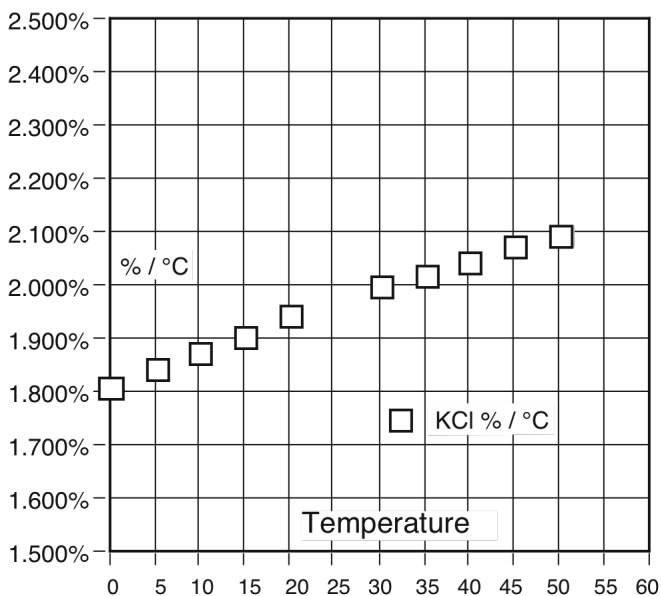


Chart 1

C. An Example of 2 different solution selections and the resulting compensation

How much error results from treating natural water as if it were KCl at 15°C?

A tap water solution should be compensated as 442™ with a tempco of 1.68 %/°C, where the KCl value used would be 1.90 %/°C.

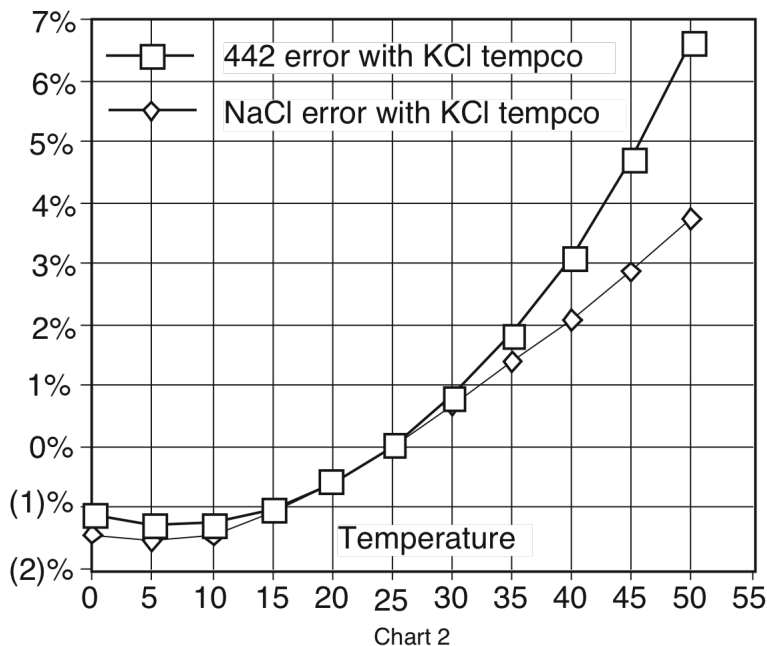
Suppose a measurement at 15°C/59°F is 900 microsiemens of true uncompensated conductivity.

Using a 442™ correction of 10 (degrees below 25) x 1.68% indicates the solution is reading 16.8% low. For correction, dividing by (.832) yields 1082 microsiemens as a compensated reading.

A KCl correction of 10 (degrees below 25) x 1.9% indicates the solution is reading 19% low. Dividing by (.81) yields 1111 microsiemens for a compensated reading. The difference is 29 out of 1082 = 2.7%.

D. A Chart of Comparative Error

In the range of 1000 μS , the error using KCl on a solution that should be compensated as NaCl or as 442TM, is illustrated in Chart 2 below.



Users wanting to measure natural water based solutions to 1% would have to alter the internal compensation to the more suitable preloaded “442TM” values or stay close to 25°C. Users who have standardized to KCl-based compensation may want to stick with it, regardless of increasing error as you get further from 25°C. The POOLPRO will provide the repeatability and convertibility of data necessary for relative values for process control.

E. Other Solutions

A salt solution like seawater or liquid fertilizer acts like NaCl. An internal correction for NaCl can be selected for greatest accuracy with such solutions.

Clearly, the solution characteristics should be chosen to truly represent the actual water under test for rated accuracy of $\pm 1\%$. Many industrial applications have historically used relative measurements seeking a number to indicate a certain setpoint or minimum concentration or trend. The POOLPRO gives the user the capability to collect data in “KCl conductivity units” to compare to older published data, in terms of NaCl or 442TM, or as appropriate. The POOLPRO can be used to reconcile data taken with other compensation assumptions.

XX. CONDUCTIVITY CONVERSION to TOTAL DISSOLVED SOLIDS (TDS)

Electrical conductivity indicates solution concentration and ionization of the dissolved material. Since temperature greatly affects ionization, conductivity measurements are temperature dependent and are normally corrected to read what they would be at 25°C (ref. Temperature Compensation, pg. 58).

A. How it's Done

Once the effect of temperature is removed, the compensated conductivity is a function of the concentration (TDS). Temperature compensation of the conductivity of a solution is performed automatically by the internal processor with data derived from chemical tables. Any dissolved salt at a known temperature has a known ratio of conductivity to concentration. Tables of conversion ratios referenced to 25°C have been published by chemists for decades.

B. Solution Characteristics

Real world applications have to measure a wide range of materials and mixtures of electrolyte solutions. To address this problem, industrial users commonly use the characteristics of a standard material as a model for their solution, such as KCl, which is favored by chemists for its stability.

Users dealing with seawater, etc., use NaCl as the model for their concentration calculations. Users dealing with freshwater work with mixtures including sulfates, carbonates and chlorides, the three predominant components (anions) in freshwater that the Myron L® Company calls “natural water”. These are modeled in a mixture called “442™” which the Myron L® Company markets for use as a calibration standard, as it does standard KCl and NaCl solutions.

The POOLPRO contains algorithms for these 3 most commonly referenced compounds. The solution type in use is displayed on the left: KCl, NaCl, and 442™.

C. When does it make a lot of difference?

First, the accuracy of temperature compensation to 25°C determines the accuracy of any TDS conversion. Assume we have industrial process water to be pretreated by RO. Assume it is 45°C and reads 1500 μ S uncompensated.

1. If NaCl compensation is used, an instrument would report 1035 μ S compensated, which corresponds to 510 ppm NaCl.

2. If 442™ compensation is used, an instrument would report 1024 μS compensated, which corresponds to 713 ppm 442™.

The difference in values is 40%.

In spite of such large error, some users will continue to take data in the NaCl mode because their previous data gathering and process monitoring was done with an older NaCl referenced device.

Selecting the correct Solution Type on the POOLPRO will allow the user to attain true TDS readings that correspond to evaporated weight.

XXI. TEMPERATURE COMPENSATION (Tempco) and TDS DERIVATION

The POOLPRO contains internal algorithms for characteristics of the 3 most commonly referenced compounds. The solution type in use is displayed on the left.

When taking conductivity measurements, the Solution Selection determines the characteristic assumed as the instrument reports what a measured conductivity would be if it were at 25°C. The characteristic is represented by the tempco, expressed in $\%/\text{°C}$. If a solution of 100 μS at 25°C increases to 122 μS at 35°C, then a 22% increase has occurred over this change of 10°C. The solution is then said to have a tempco of 2.2 $\%/\text{°C}$.

Tempco always varies among solutions because it is dependent on their individual ionization activity, temperature and concentration. This is why the POOLPRO features mathematically generated models for known salt characteristics that also vary with concentration and temperature.

XXII. pH and ORP/FC^E

A. pH

1. pH as an Indicator

pH is the measurement of Acidity or Alkalinity of an aqueous solution. It is also stated as the Hydrogen Ion activity of a solution. pH measures the effective, not the total, acidity of a solution.

A 4% solution of acetic acid (pH 4, vinegar) can be quite palatable, but a 4% solution of sulfuric acid (pH 0) is a violent poison. pH provides the needed quantitative information by expressing the degree of activity of an acid or base.

In a solution of one known component, pH will indicate concentration indirectly. However, very dilute solutions may be very slow reading, just because the very few ions take time to accumulate.

2. pH Units

The acidity or alkalinity of a solution is a measurement of the relative availabilities of hydrogen (H^+) and hydroxide (OH^-) ions. An increase in (H^+) ions increases acidity, while an increase in (OH^-) ions increases alkalinity. The total concentration of ions is fixed as a characteristic of water, and balance would be 10^{-7} mol/liter (H^+) and (OH^-) ions in a neutral solution (where pH sensors give 0 voltage).

pH is defined as the negative logarithm of hydrogen ion concentration. Where (H^+) concentration falls below 10^{-7} , solutions are less acidic than neutral, and therefore are alkaline. A concentration of 10^{-9} mol/liter of (H^+) would have 100 times less (H^+) ions than (OH^-) ions and be called an alkaline solution of pH 9.

3. The pH Sensor

The active part of the pH sensor is a thin glass surface that is selectively receptive to hydrogen ions. Available hydrogen ions in a solution will accumulate on this surface and a charge will build up across the glass interface. The voltage can be measured with a very high impedance voltmeter circuit; the dilemma is to connect the voltmeter to solution on each side.

The glass surface encloses a captured solution of potassium chloride holding an electrode of silver wire coated with silver chloride. This is the most inert connection possible from a metal to an electrolyte. It can still produce an offset voltage, but using the same materials to connect to the solution on the other side of the membrane causes the 2 equal offsets to cancel.

The problem is, on the other side of the membrane is an unknown test solution, not potassium chloride. The outside electrode, also called the Reference Junction, is of the same construction with a porous plug in place of a glass barrier to allow the junction fluid to contact the test solution without significant migration of liquids through the plug material. Figure 28 shows a typical 2 component pair. Migration does occur, and this limits the lifetime of a pH junction from depletion of solution inside the reference junction or from contamination. The junction may be damaged if dried out because insoluble crystals may form in a layer, obstructing contact with test solutions.

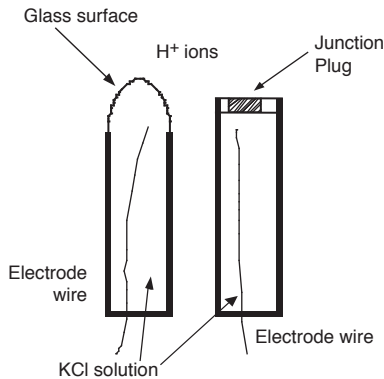


Figure 28

4. The Myron L® Integral pH Sensor

The sensor in the POOLPRO (see Figure 29) is a single construction in an easily replaceable package. The sensor body holds an oversize solution supply for long life. The reference junction “wick” is porous to provide a very stable, low permeable interface, and is located under the glass pH sensing electrode. This construction combines all the best features of any pH sensor known.

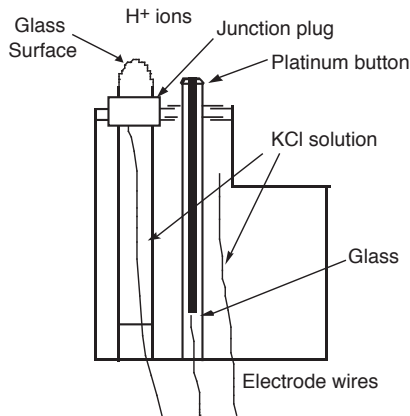


Figure 29

5. Sources of Error

The basics are presented in pH and ORP/FC^E, pg. 58.

a. Reference Junction

The most common sensor problem will be a clogged junction because a sensor was allowed to dry out. The symptom is a drift in the “zero” setting at 7 pH. This is why the POOLPRO PS9 does not allow more than 1 pH unit of offset during calibration. At that point the junction is unreliable.

b. Sensitivity Problems

Sensitivity is the receptiveness of the glass surface. A film on the surface can diminish sensitivity and cause a long response time.

c. Temperature Compensation

pH sensor glass changes its sensitivity slightly with temperature, so the further from pH 7 one is, the more effect will be seen. A pH of 11 at 40°C would be off by 0.2 units. The POOLPRO PS9 senses the sensor well temperature and compensates the reading.

B. ORP/Oxidation-Reduction Potential/REDOX

1. ORP as an Indicator

ORP is the measurement of the ratio of oxidizing activity to reducing activity in a solution. It is the potential of a solution to give up electrons (oxidize other things) or gain electrons (reduce).

Like acidity and alkalinity, the increase of one is at the expense of the other, so a single voltage is called the Oxidation-Reduction Potential, with a positive voltage showing, a solution wants to steal electrons (oxidizing agent). For instance, chlorinated water will show a positive ORP value.

2. ORP Units

ORP is measured in millivolts, with no correction for solution temperature. Like pH, it is not a measurement of concentration directly, but of activity level. In a solution of only one active component, ORP indicates concentration. Also, as with pH, a very dilute solution will take time to accumulate a readable charge.

3. ORP Sensors

An ORP sensor uses a small platinum surface to accumulate charge without reacting chemically. That charge is measured relative to the solution, so the solution "ground" voltage comes from a reference junction - same as the pH sensor uses.

4. The Myron L® ORP Sensor

Figure 29, pg. 60, shows the platinum button in a glass sleeve. The same reference is used for both the pH and the ORP sensors. Both pH and ORP will indicate 0 for a neutral solution. Calibration at zero compensates for error in the reference junction.

A zero calibration solution for ORP is not practical, so the POOLPRO uses the offset value determined during calibration to 7 in pH calibration (pH 7 = 0 mV). Sensitivity of the ORP surface is fixed, so there is no gain adjustment either.

5. Sources of Error

The basics are presented in pH and ORP/FC^E, pg. 58, because sources of error are much the same as for pH. The junction side is the same, and though the platinum surface will not break like the glass pH surface, its protective glass sleeve can be broken. A surface film will slow the response time and diminish sensitivity. It can be cleaned off with detergent or acid, as with the pH glass.

C. Free Chlorine Equivalent

1. FC^E as an Indicator of Sanitizing Strength

Chlorine, which kills bacteria by way of its power as an oxidizing agent, is the most popular germicide used in water treatment. Chlorine is not only used as a primary disinfectant, but also to establish a sufficient residual level of Free Available Chlorine (FAC) for ongoing disinfection.

FAC is the chlorine that remains after a certain amount is consumed by killing bacteria or reacting with other organic (ammonia, fecal matter) or inorganic (metals, dissolved CO₂, Carbonates, etc) chemicals in solution. Measuring the amount of residual free chlorine in treated water is a well accepted method for determining its effectiveness in microbial control.

The Myron L[®] Company FC^E method for measuring residual disinfecting power is based on ORP, the specific chemical attribute of chlorine (and other oxidizing germicides) that kills bacteria and microbes.

2. FC^E Units

The PS9 is the first handheld device to detect free chlorine directly, by measuring ORP. The ORP value is converted to a concentration reading (ppm) using a conversion table developed by Myron L[®] Company through a series of experiments that precisely controlled chlorine levels and excluded interferants.

Other test methods typically rely on the user visually or digitally interpreting a color change resulting from an added reagent-dye. The reagent used radically alters the samples pH and converts the various chlorine species present into a single, easily measured species. This ignores the effect of changing pH on free chlorine effectiveness and disregards the fact that some chlorine species are better or worse sanitizers than others.

The Myron L[®] Company PS9 avoids these pitfalls. The chemistry of the test sample is left unchanged from the source water. It accounts for the effect of pH on chlorine effectiveness by including pH in its calculation. For these reasons, the POOLPRO's FC^E feature provides the best reading-to-reading picture of the rise and fall in sanitizing effectivity of free available chlorine.

The PS9 also avoids a common undesirable characteristic of other ORP-based methods by including a unique Predictive ORP value in its FC^E calculation. This feature, based on a proprietary model for ORP sensor behavior, calculates a final stabilized ORP value in 1 to 2 minutes rather than the 10 to 15 minutes or more that is typically required for an ORP measurement.

XXIII. ALKALINITY, HARDNESS AND LSI FUNCTIONS

A. Alkalinity and Hardness Titrations

The PS9 uses a conductometric method of titration. This means that instead of measuring a color or pH change, it measures the change in conductivity when reagent is added. The endpoint is determined based on the conductivity changes and then the alkalinity/hardness is calculated based on the total volume of titrant consumed. For hardness titration, the sample conductivity must be $\leq 10,000$ uS/cm in KCl mode. It may not work with strong buffered samples or samples with extreme pH (e.g. < 2 pH). The titration is based on EDTA complexation reactions. Metal ions from the alkaline earth metal group, such as Ba^{2+} and Sr^{2+} will be measured as hardness. Interferences including aluminum, iron, nickel, cobalt, manganese, etc. may be corrected by inhibitors.

B. Langelier Saturation Index

The LSI calculator function uses a calcium carbonate saturation index algorithm developed by Dr. Wilfred Langelier in 1936:

$$\text{SI} = \text{PH} + \text{TF} + \text{CF} + \text{AF} - 12.1$$

Where:

PH = pH value

TF = $0.0117 \times \text{Temperature value} - 0.4116$

CF = $0.4341 \times \ln(\text{Hardness value}) - 0.3926$

AF = $0.4341 \times \ln(\text{Alkalinity value}) - 0.0074$

The following is a general industry guideline for interpreting LSI values:

- An index between -0.5 and +0.5 indicates balanced water
- An index of more than +0.5 indicates scale-forming water
- An index below -0.5 indicates corrosive water

You can use the calculator to adjust any of the water balance parameter variables used in the calculation to analyze the effect of the change on water balance.

The LSI calculator function generates an accurate saturation index value that indicates the state of the system at the time the calculation is made.

C. Hardness Units

NOTE: The PS9 MUST be in the HARDNESS FUNCTION for hardness unit preference selection.

The hardness unit preference is set simultaneously for both functions that capture a hardness value. That means when you set the unit preference in Hardness Function, it is set for the LSI Calculator function at the same time.


The LSI Calculator allows you to calculate with either ppm or grains as hardness units.

The hardness unit conversion is based on the following equivalency:

$$17.1 \text{ ppm (mg/L)} = 1 \text{ grain}$$

XXIV. SOFTWARE VERSION

Contact the Myron L[®] Company to see if a software upgrade is available.

1. Press any parameter key.
2. Press  key until three numbers are displayed as shown in Figure 30.

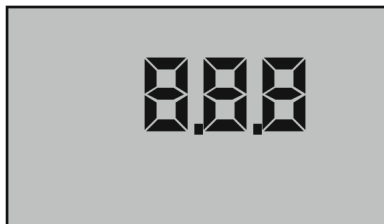


Figure 30

3. Press any parameter key; instrument will time out in ~15 seconds.

XXV. GLOSSARY

Anions	Negatively charged ions. See Solution Characteristics, pg. 57.
Algorithm	A procedure for solving a mathematical problem. See Temperature Compensation (Tempco) and TDS Derivation, pg. 58.
FAC	Free Available Chlorine. The amount of chlorine that remains active in solution and is available for ongoing disinfection. See Free Chlorine as an Indicator, pg. 62.
FC^E	FC ^{E™} directly measures ORP, the germ killing property of chlorine and other oxidizing germicides. It displays both the ORP reading (in mVDC) as well as an equivalent free chlorine concentration (in familiar ppm). For more information see <i>FC^{E™}: Groundbreaking Measurement of Free Chlorine Disinfecting Power in a Hand-Held Instrument</i> on the Myron L [®] Company website.
Logarithm	An arithmetic function. The inverse of an exponential function. See pH Units, pg. 59.
ORP	Oxidation-Reduction Potential or REDOX, See ORP/ Oxidation-Reduction Potential/REDOX, pg. 61.
REDOX Reaction	An abbreviation for Reduction-Oxidation reactions. This is the basic electrochemical process by which chlorine destroys microbes by grabbing electrons from the microbe's proteins, denaturing the protein and killing the organism. ORP directly measures the strength of a solutions' REDOX potential and, therefore, sanitizing strength.
TDS	Total Dissolved Solids or the Total Conductive Ions in a solution. See Conductivity Conversion to Total Dissolved Solids (TDS), pg. 58.
Tempco	Temperature Compensation See Temperature Compensation of Aqueous Solutions, pg. 54.

For details on specific areas of interest refer to the Table of Contents, pg. 5.

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